

Nº 174103

**Project: Nd/Di production from CBMM Didymium oxide**

**João Batista Ferreira Neto.**

*Palestra apresentada no Workshop Brasil-Alemanha em Terras Raras no IPT, São Paulo, 2016*

A série “Comunicação Técnica” compreende trabalhos elaborados por técnicos do IPT, apresentados em eventos, publicados em revistas especializadas ou quando seu conteúdo apresentar relevância pública.

# Summary

- IPT and Center for Technology in Metallurgy and Materials
  - Laboratory of Metallurgical Processes
    - Examples of R&D&I projects
- Project: Didymium production from CBMM oxide
  - Next step..... Alloy and magnet production
- Possibilities of Cooperation with German Institutions and Universities



Institute for Technological Research

Technological solutions

# Who we are

- Institute for Technological Research of the State of São Paulo S.A.
- One of the first applied R&D&I institutions in Brazil and the largest applied multipurpose R&D&I institution in Latin America
- Belongs to the Development Secretariat of the State of São Paulo
- IPT provides technological solutions to public and private companies and institutions



# Where we are

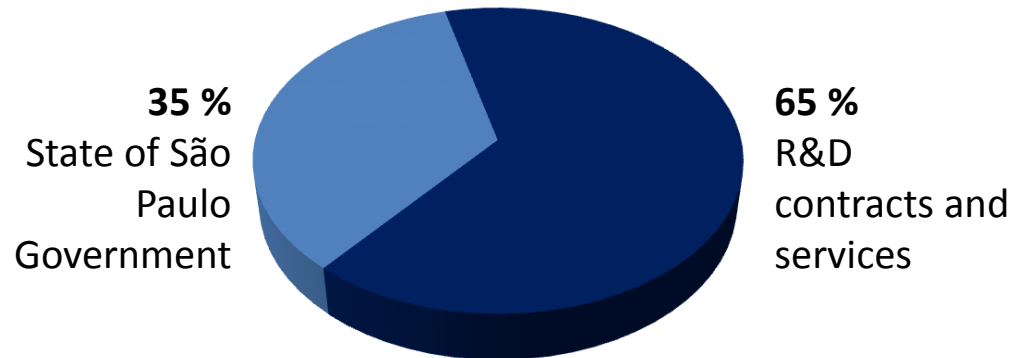
## Units

- São Paulo
- São José dos Campos (composite materials)
- Franca (individual protection equipment)



# 2015 Figures

- Annual revenue: **US\$ 45 million**



## Technical Production:

- Technical Documents: **22,661** technical reports
- Number of clients: **3,794**
- Ongoing projects: **145**
- Publications: **302** papers in journals and congresses
- Patents and softwares: **10**

## Human Resources Dec. 2015

Researchers	381
Technicians	196
Administrative Support	250
Interns	80
Total	907

1.00 USD = 3,9042 BRL

# Technical Activities 2015



## **Innovation, research and development**

31% of the total revenue

**(TARGET : 40% IN 2018)**



## **Technological Services**

26,6% of the total revenue



## **Development and metrological support**

41,9% of the total revenue



## **Information and technology education**

0,5% of the total revenue

# Markets

## Energy

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**Oil and gas**  
**Ethanol**

## Transportation

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**Roads**  
**Naval**  
**Pipeline**  
**Metro-railway**  
**Airspace**  
**Cargo**  
**IT & ITS**

## Materials and Chemistry

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**Metallurgy**  
**Chemistry**  
**Bioproducts**  
**Plastics & Rubber**  
**Composites**  
**Textiles & Leather**  
**Wood**

## Infrastructure

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**Civil Works**  
**Buildings**  
**Environmental Impacts**  
**Mining**

# Technical Centers

■ CTGeo  
Center for  
Geoenvironmental  
Technologies

■ CETAC  
Center for the Built  
Environment

■ CT-Obras  
Center for  
Infrastructure Work  
Technology

■ NT – MPE  
Nucleus for  
Technological Support  
to Medium and to  
Small Enterprises

■ CT-Floresta  
Center for Forest  
Resource Technology

■ CQuiM  
Center for Chemistry and  
Manufactured Goods

■ CIAM  
Center for Information  
Technology, Automation  
and Mobility

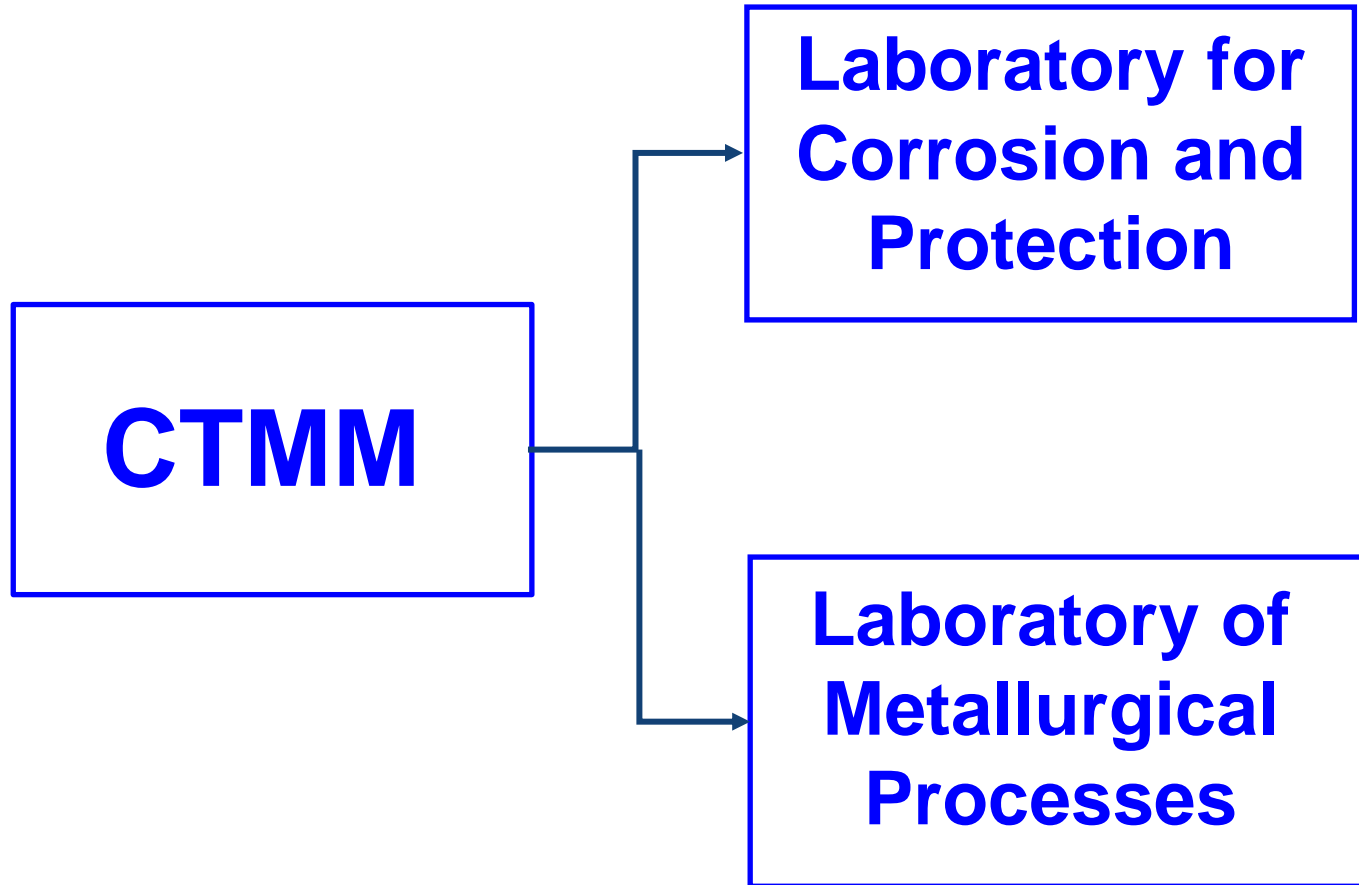
■ NT- BIONANO  
Nucleus for  
Bionanomanufacturing

■ CTMM  
Center for Technology  
in Metallurgy and  
Materials

■ CTMetro  
Center for Mechanical,  
Electrical and Fluid Flow  
Metrology

■ CTMNE  
Center for Mechanical,  
Naval and Electrical  
Technologies

■ LEL  
Light Weight Laboratory



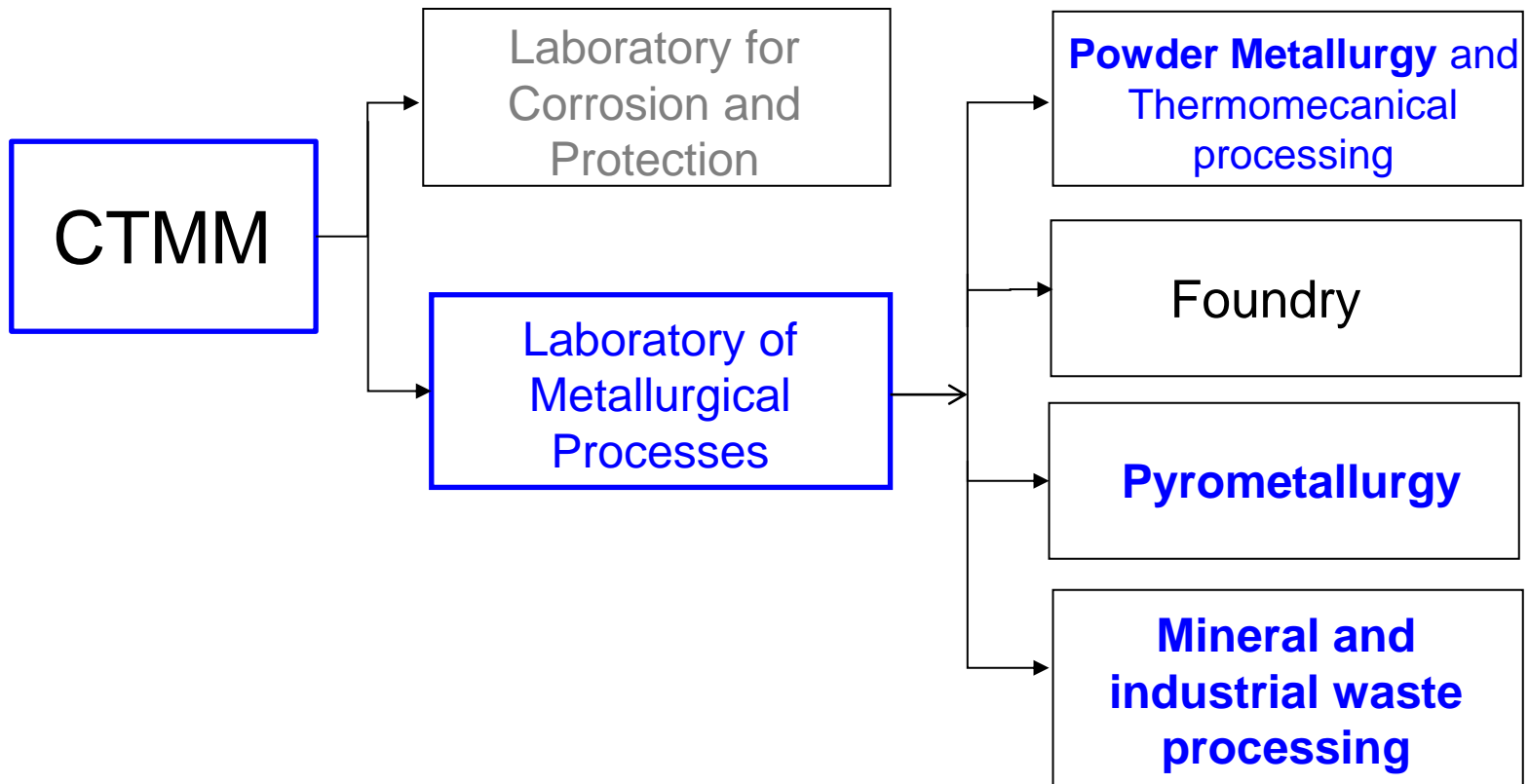
# CTMM-Figures

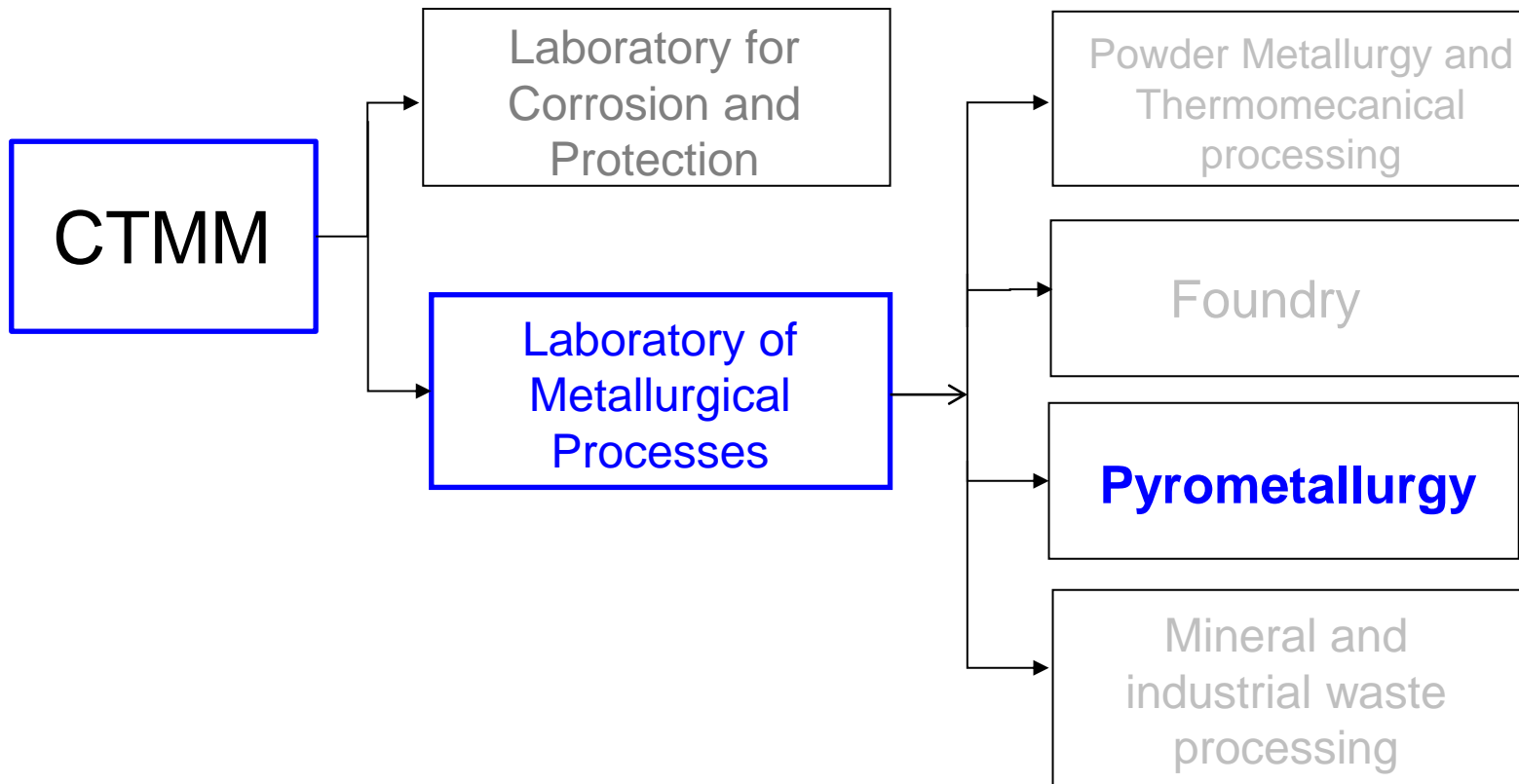
<b>Human Resources Dec. 2015</b>	
PhD	12
MSc	15
Undergraduate	11
Technicians	35
Administrative	4
Interns	18
<b>Total</b>	<b>95</b>



ANNUAL BUDGET (2015)  
US\$ 3.5 million

1.00 USD = 3.9042 BRL





# Laboratory of Metallurgical Processes - LPM

## Experimental Facilities

- Induction Furnaces (air and vacuum): 15 kW (5 kg) - 135 kW (500 kg)
- Vacuum Induction Furnace with directional solidification device (5kg; 20 kW)



- Electroslag Remelting - ESR ( $\phi$  6" x 800 mm) – composite rolling mill – alloy of core distinct from alloy of surface
- Graphite Resistance Furnace (up to 2000° C) controlled atmosphere
- Experimental Rolling mill

# Laboratory of Metallurgical Processes - LPM

## Experimental Facilities

- EAF - 400 kW (1 ton)



- Three phase EAF
- System for oxygen and argon injection
- Temperature control
- Celox (Slac – %FeO) and oxygen activity in steel bath
- Bag house

- Multicrystallizer Furnace – multicrystalline silicon growth (20 kg and 90 kg)



- Steel centrifugation (100 kg steel)

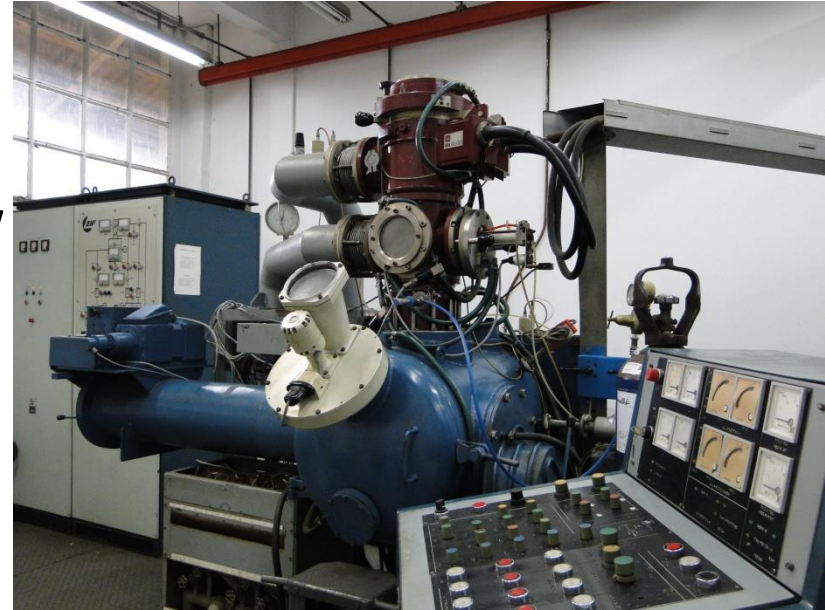


- Submerged Electric Arc Furnace - 30 KVA

# Laboratory of Metallurgical Processes - LPM

## Experimental Facilities

- Electron Beam Furnace (EB) - 80 kW



- Laboratory Rotary Kiln Furnace (up to 1500° C)



# Laboratory of Metallurgical Processes - LPM

## Experimental Facilities

### Equipments for special experiments:

- Differential Thermoanalysis and Thermogravimetric (DTA, DSC, TG) up to 1600° C or 2000° C.
- Processes gases analyses by chromatography and mass spectrometry.

### Chemical Analyses

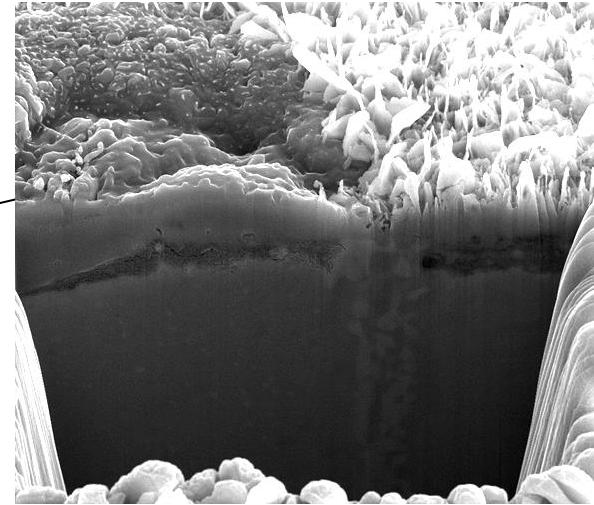
- Clean room – ISO Class 8
- Spectrophotometry of Atomic Absorption
- LECO (C, S, O, N, H).
- ICP e X-Ray Fluorescence
- XRD
- Microstructural characterization  
Optical and SEM (WDS-EDS)



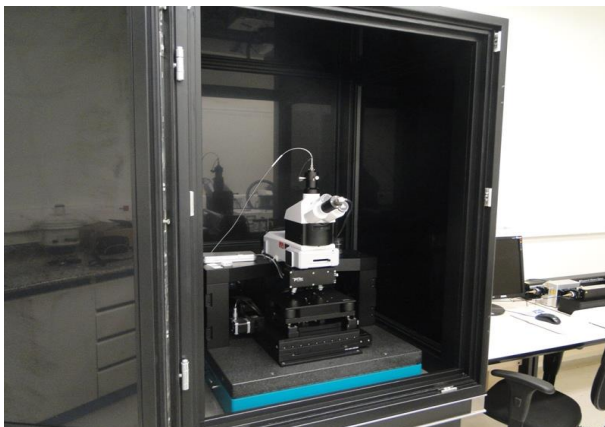
# Characterization techniques



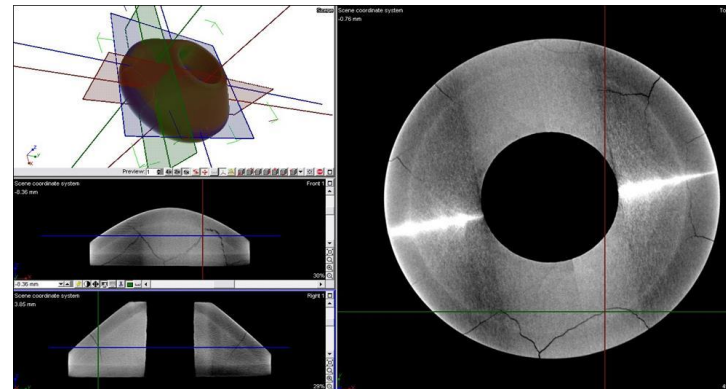
**SEM FEG/FIB**



**Oxidized HSS for rolls  
(surface and cross section)**



**Raman spectroscopy**



**X-ray  
microtomography**



# Pyrometallurgy

## Main Activities


- Melting and Refining of Special Materials
- Carbothermic and Metalothermic reduction of Slags, Ores and Concentrates
- Modeling and determination of thermodynamic parameters of Metallurgical Processes

# Pyrometallurgy

## Examples of ongoing and developed projects

- Inclusions controlling in steels
- Metallurgical route for solar grade silicon production 
- Ni recovery from scraps and metallurgical by-products
- Titaniferous slag production by ilmenite concentrate reduction in self-reduction pellets
- Modification of molten Steel slag (BOF) for cement application 
- DeP of Mn steels and stainless steels under reducing conditions (P removed as a Phosphide)

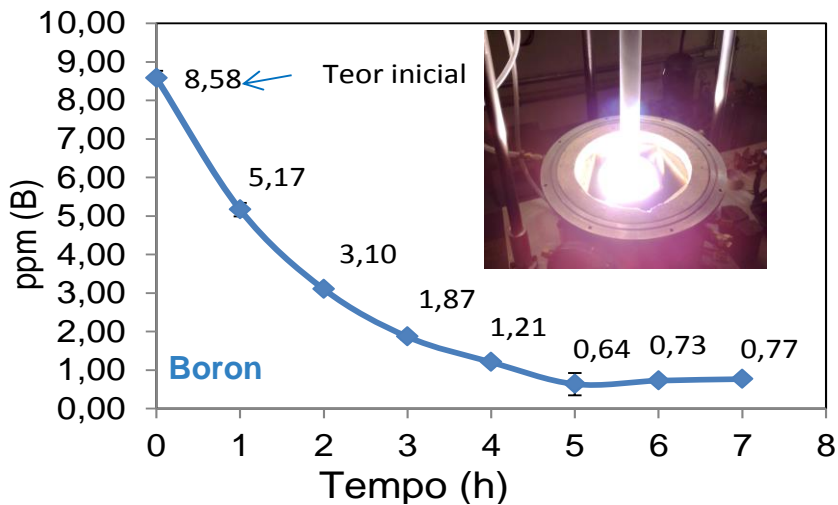
## Examples of ongoing and developed projects

- Development of the route of Nb concentrate refining by selective carbothermic reduction – CBMM 
- Study of aluminothermic reduction of liquid niobium concentrate
- Evaluation of the Fe-Nb dissolution in cast irons and steels (effect of particle size and Temperature on the dissolution kinetic)
- Thermodynamic Modelling and simulation (FactSage and Thermocalc):
  - Evaluation of inclusions formation in steels
  - Simulation of reduction of the electric arc furnace dust (Zn recovery)
  - Simulation of Metalothermic reduction of oxygen steelmaking slags
  - Simulation of Hydrogen removal from steel during ladle vacuum treatment

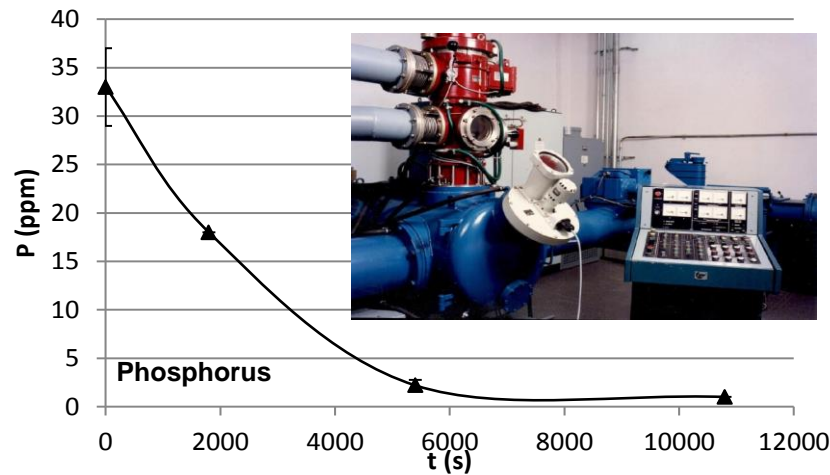


# Pyrometallurgical refining – SG Silicon

Plasma/IF furnace

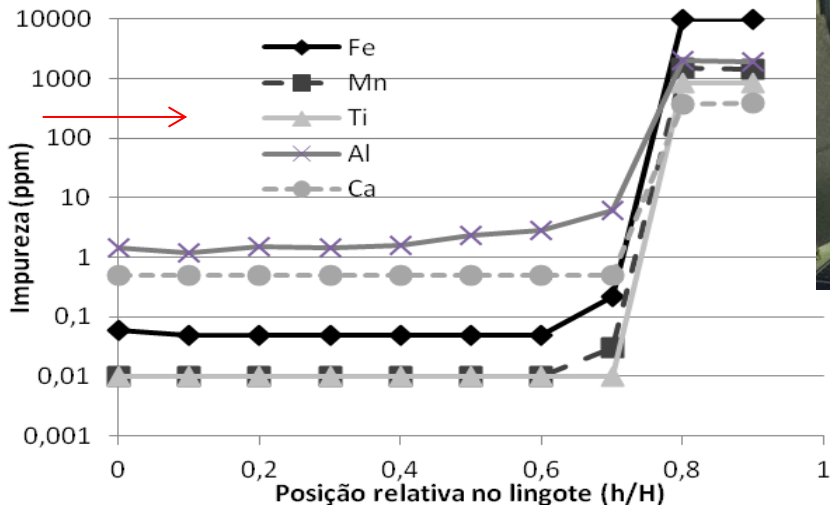


EB furnace



→  
**MG Silicon** → **SG Silicon**  
 98%-99% → > 99,999%

Controlled solidification



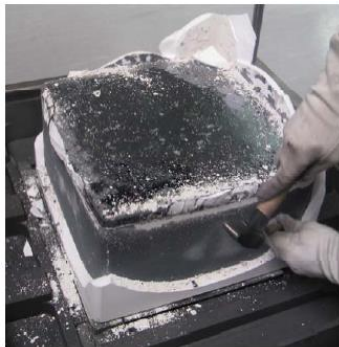
# Pyrometallurgical refining – SG Silicon



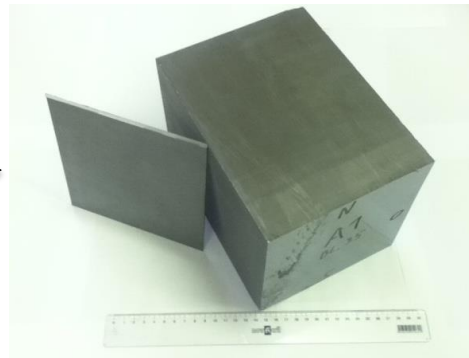
**SG Silicon  
(20 kg)**



**Directional  
solidification  
furnace**



**SG multicrystalline  
Silicon**



**Columnar grains**



# Blast Furnace slag X Steel Slag → Cement Industry in Brazil

- Near future scenario: There will be a shortage of BF slag to supply the demand of cement manufacturing in Brazil.
- The steel production in Brazil has been around 32-34 millions t/y for last ten years. There is no expectation of increasing X growth of construction industry in Brazil.
- Steel slag could be an alternative as cement mineral admixture, partially substituting the BF slag.

**Objective: Development of autogenous process of liquid Steel Slag modification aiming its application as raw material in the portland cement manufacturing, partially substituting the BF slag**

## **Technological barriers to be overcome:**

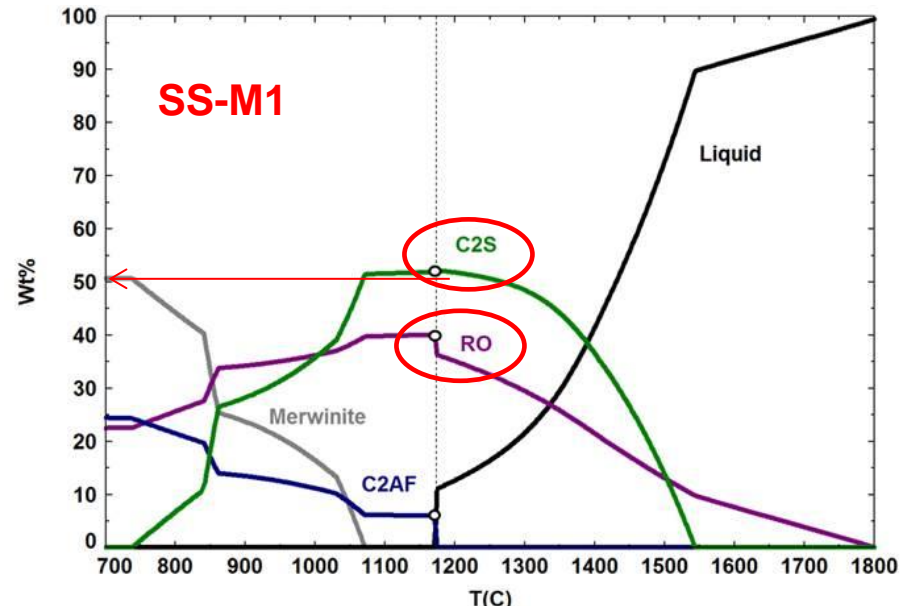
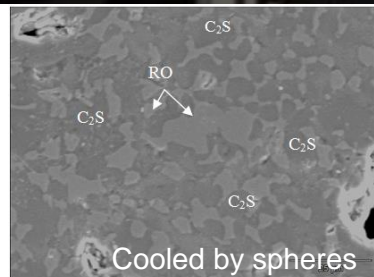
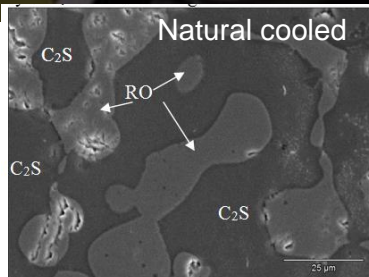
- Chemical composition modification (decreasing of free CaO and MgO, Fe<sub>2</sub>O<sub>3</sub>/FeO and Fe)
- Transformation of mineralogical phases (appropriate phases with hydraulic activity for cement production)
- Low cost by-products/residues must be used as transforming agents (achieve economic feasibility)
- Use of heat content in SSlag for modification process

# Effect of cooling rate and chemical composition on slags crystallization

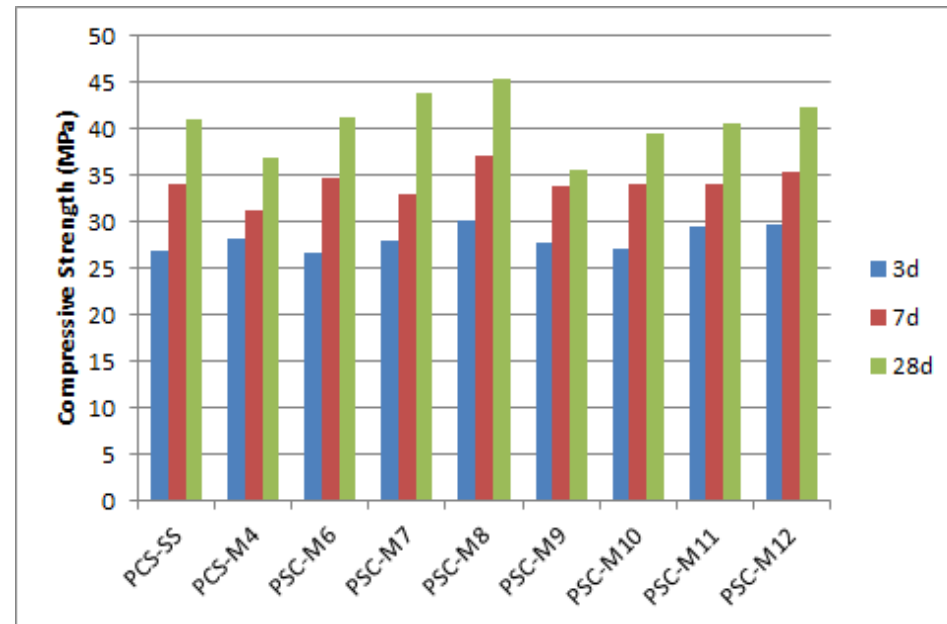
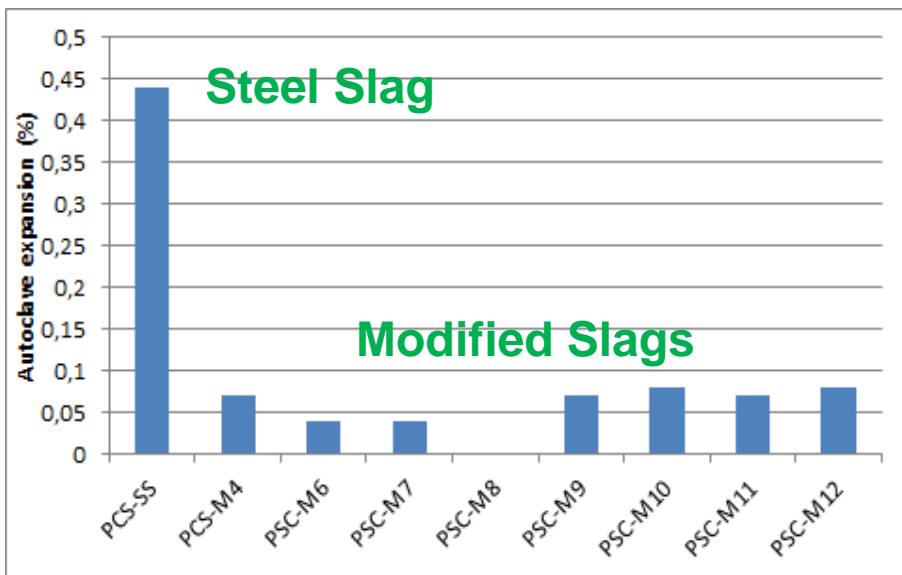
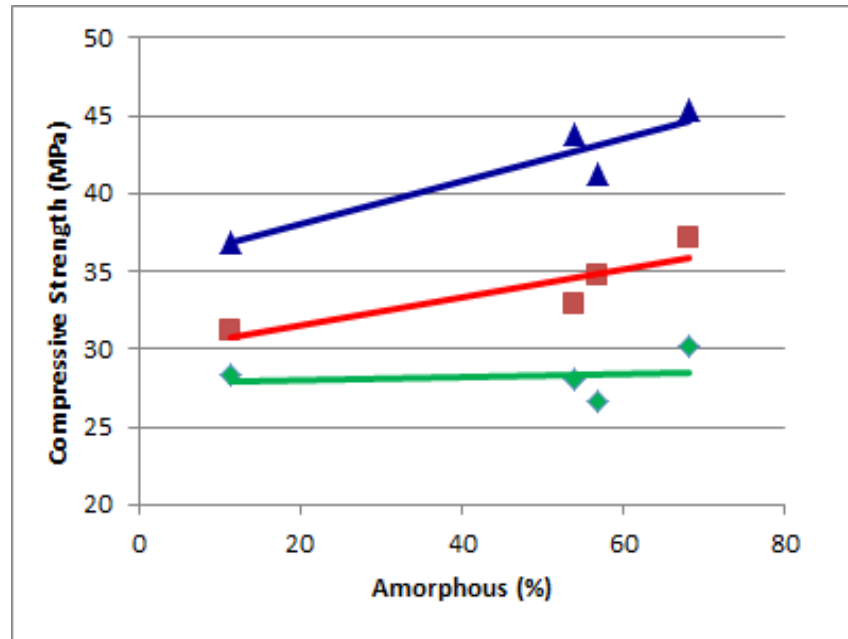
## Pilot Scale (300 kg modified slag)



Cooling Simulation

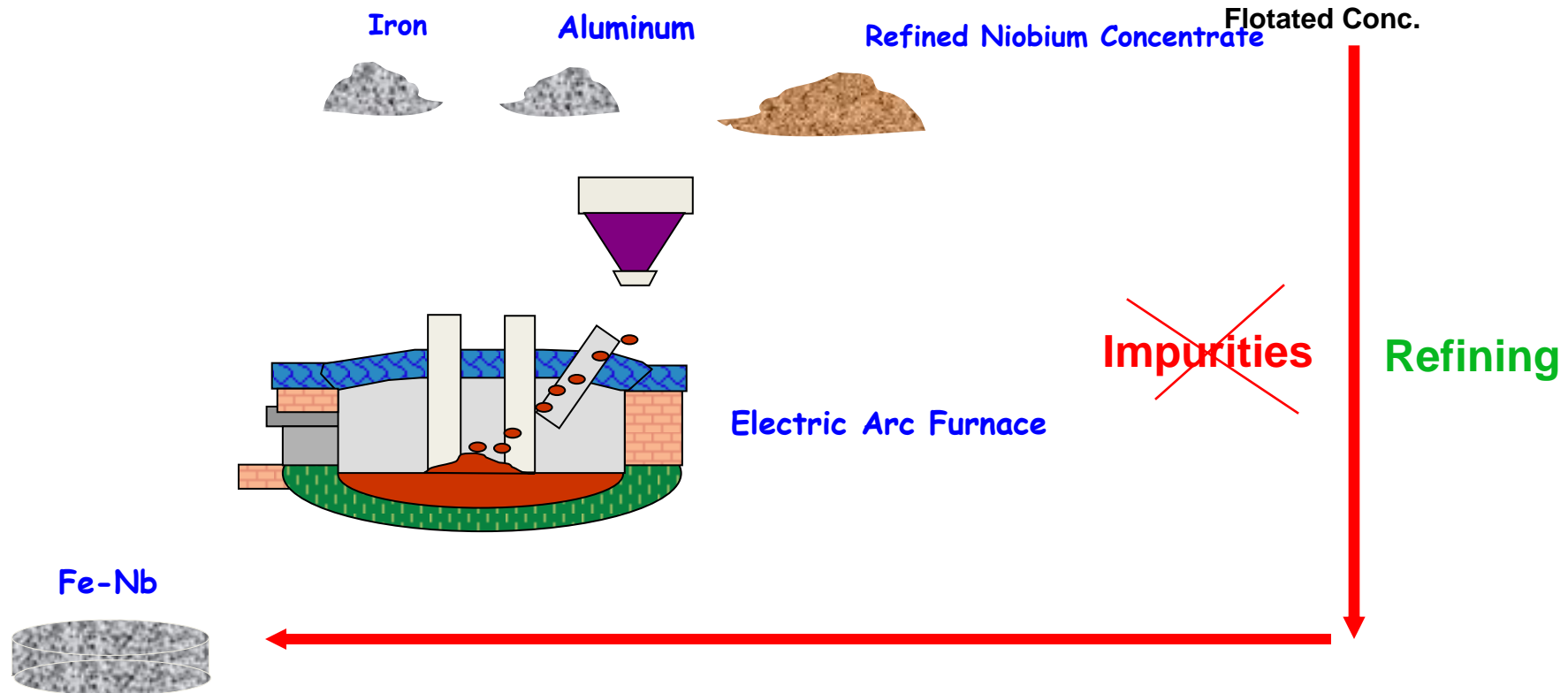


# 75% of ordinary Portland cement + 25% of slag (SS and modified slags)



# Refining of Nb concentrate by selective carbothermic reduction

Nb <sub>2</sub> O <sub>5</sub> (%)	CaO (%)	<b>P</b> (%)	<b>Pb</b> (%)	SiO <sub>2</sub> (%)	BaO (%)	ThO <sub>2</sub> (%)	<b>Sn</b> (%)	<b>Fe<sub>2</sub>O<sub>3</sub></b> (%)	S (%)	H <sub>2</sub> O (%)
55-60	0-0,2	<b>0,3-0,8</b>	<b>0,2-1</b>	0,1-0,5	15-18	2-3	<b>0,1</b>	<b>2-8</b>	0,02-2	5-7



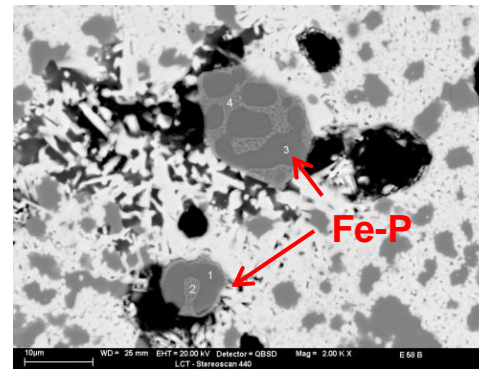
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## Refining

Carbothermic  
reduction of Nb  
concentrate

Reduction of P, Pb, Sn,  
(Fe)  
Prevent Nb reduction



Conc. Flotado.

Refined  
concentrate

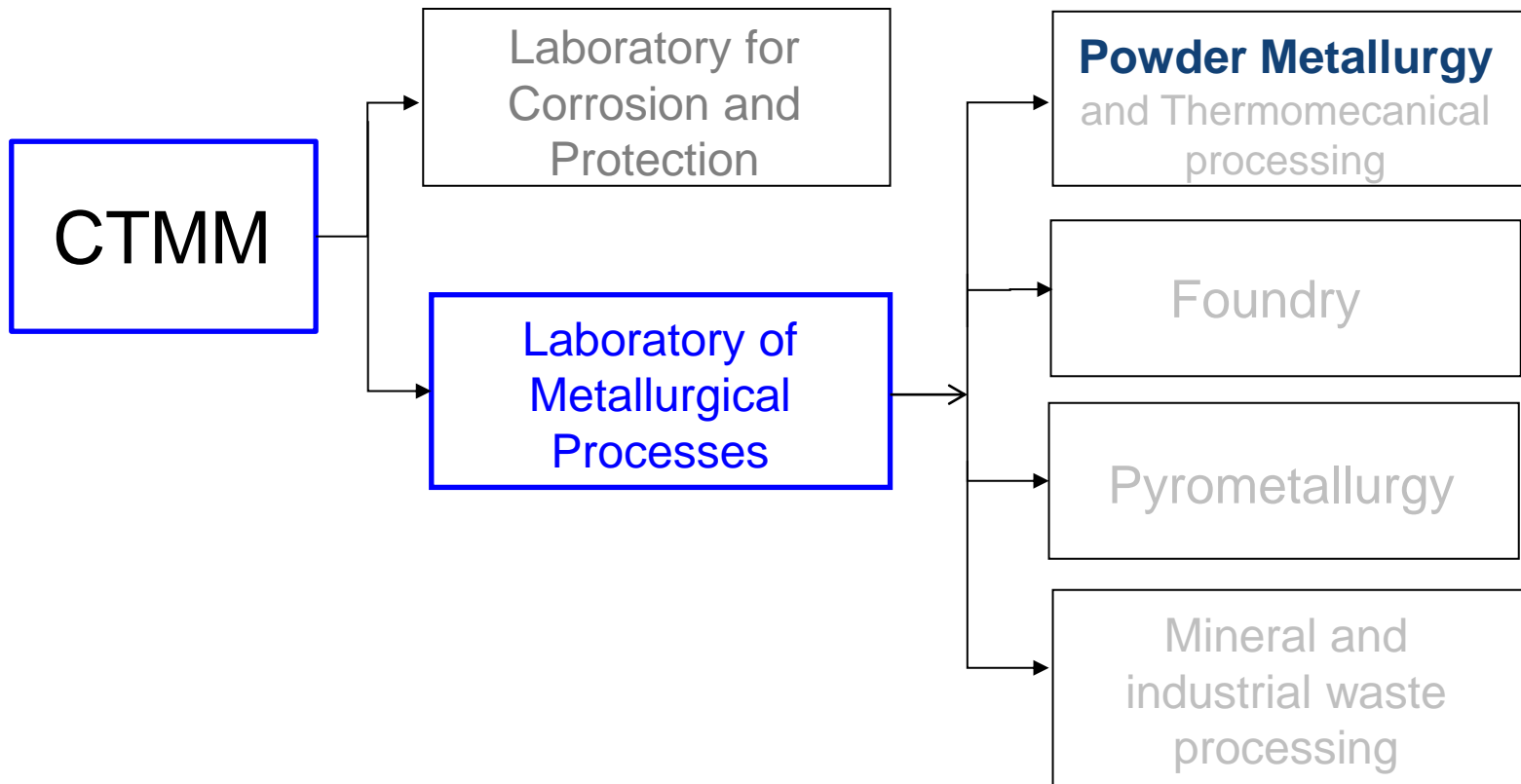
Fe-P-Sn; Pb

Melting and metallic phase  
(Fe-P alloy) separation from  
refined concentrate

Pilot scale (300 kg by batch) in EAF  
> 40 t of Nb conc. refined at IPT

Process adopted by CBMM since 2000  
Capacity ~ 80.000 t conc DeP/y





# Laboratory of Metallurgical Processes - LPM

## Experimental Facilities

- Powders production
  - Powder atomizers (water and inert gas)

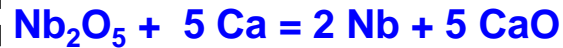
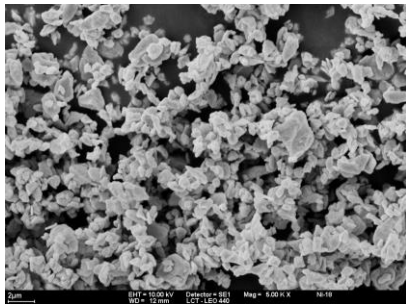


# Nb and NbO powder for capacitors

Hydriding-Milling-  
Dehydriding

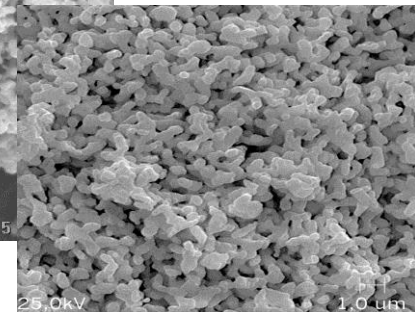
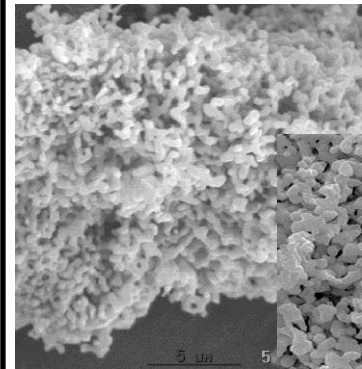
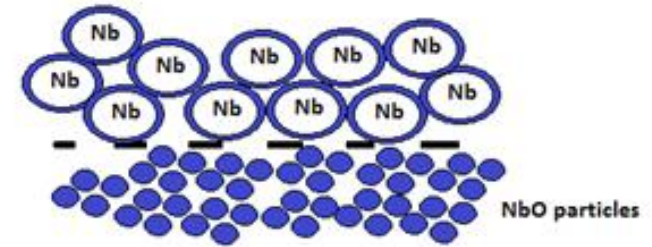
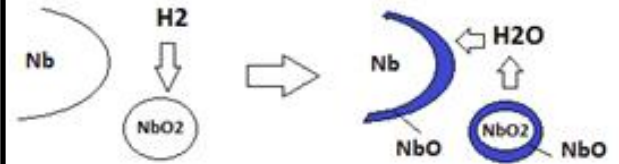
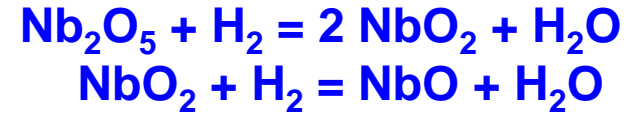
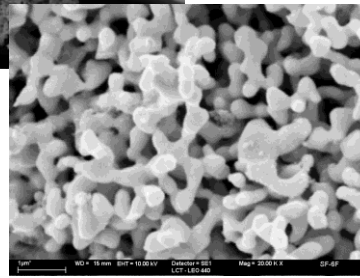
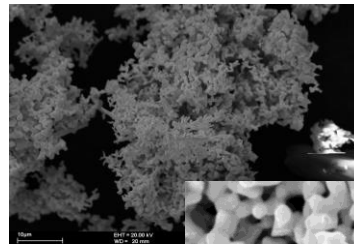
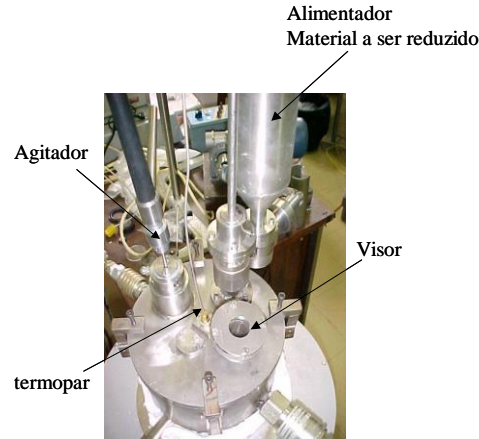


Nb (100 kg)  
after Hydriding



**Molten Salt: CaCl<sub>2</sub>**

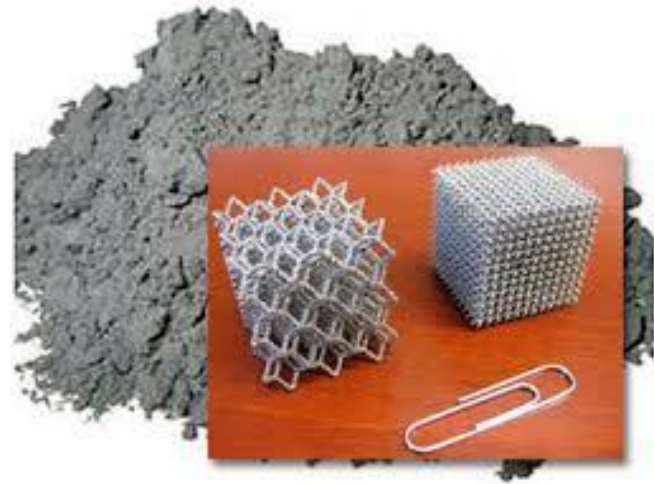
**High Ca and CaO solubility**

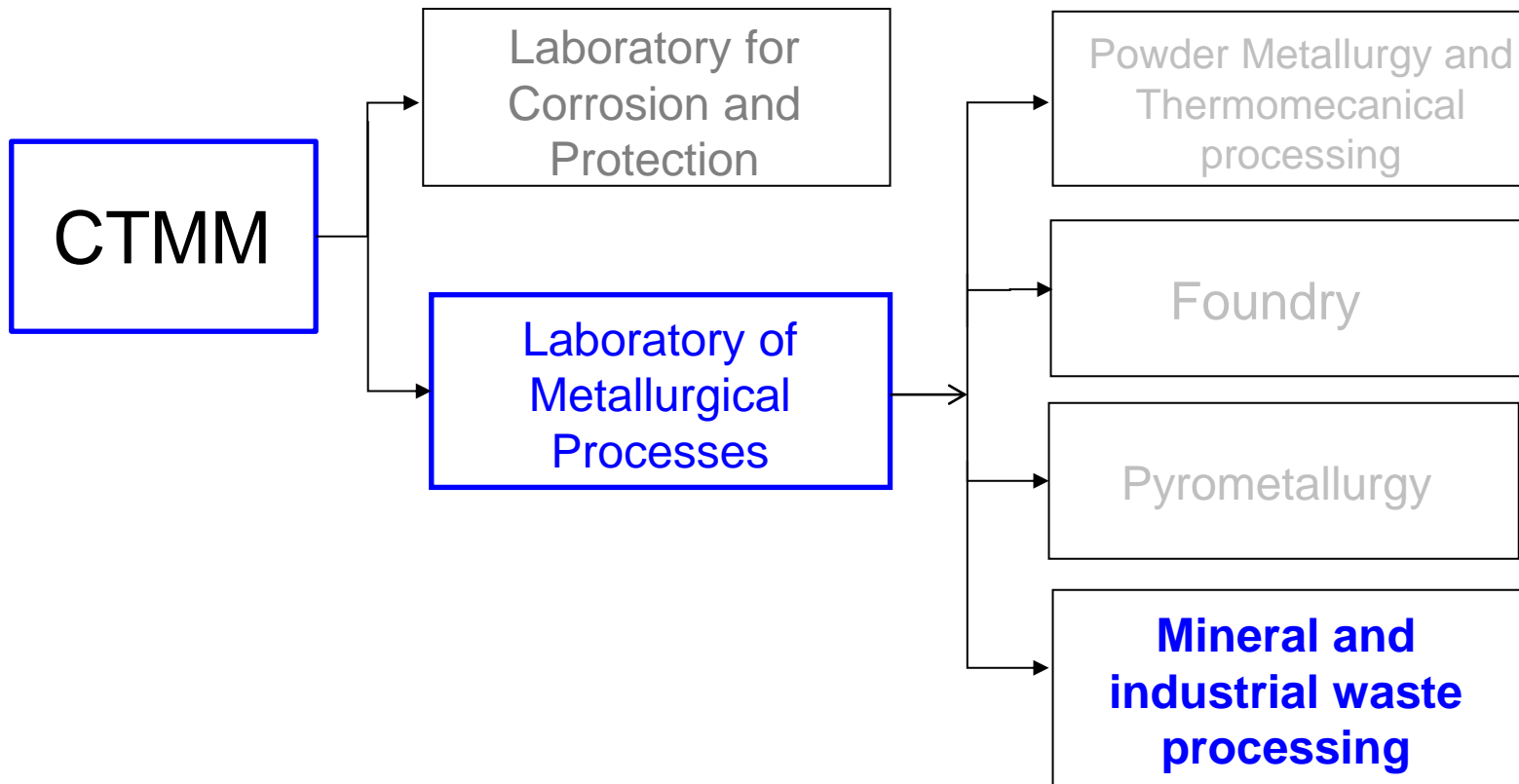


Sample	Capacitance (kCV/g)	DCL (nA/CV)
Market specification	85 ~ 105	< 0,35
Pilot plant	112 ~ 140	0,12 ~ 0,28

# Ti-Nb-Zr and Nb-Ti powders for Additive Manufacturing

- **Orthopedic Implants**
- **Alloys melting in a EB furnace**
- **Powders production:**
  - **Inert gas atomization**
  - **H-DH + Milling**
  - **H-DH + Milling + plasma spherodizing**
- **Implants produced by additive manufacturing**
- **Biocompatibility tests**





# Mineral and Industrial Waste Processing

- Expertise in applying mineral processing techniques
  - Ores
  - Fertilizers
  - Industrial wastes (metallurgy, construction, agroindustrial, etc)
- Batch and pilot scales equipment
  - 450 m<sup>2</sup>

# Mineral and Industrial Waste Processing

## Mineral Processing

- Comminution processes (batch / pilot )
  - Evaluation of grinding media wear
  - Assessment of chemical supplies for grinding process
  - Determination of process parameters (grinding and classification)



# Mineral and Industrial Waste Processing

## Mineral Processing

### *Enrichment processes*

### *(ores and tailings)*

- Density separation
  - Jig
  - Shaking table
  - Heavy liquid



- Flotation
  - Kinetic
  - Evaluation of chemical supplies

# Mineral and Industrial Waste Processing

## Mineral Processing

### *Enrichment processes*

- Electrostatic separation
- Magnetic separation



- Classification
  - Cyclone
  - Spiral classifier
  - Air classifier

# Mineral and Industrial Waste Processing

## Fine particles agglomeration processes

- Binders exploration
- Process development (pelletization and briquetting)
- Biomass: physical pre-treatment and densification



# Project: Nd/Di production from CBMM Didymium oxide



Next step..... Alloy and magnet  
production

## World Reserves of RE

	Mine production <sup>e</sup>		Reserves <sup>5</sup>
	<u>2013</u>	<u>2014</u>	
United States	5,500	7,000	1,800,000
Australia	2,000	2,500	<sup>6</sup> 3,200,000
Brazil	330	—	22,000,000
China	95,000	95,000	55,000,000
India	2,900	3,000	3,100,000
Malaysia	180	200	30,000
Russia	2,500	2,500	(7)
Thailand	800	1,100	NA
Vietnam	220	200	(7)
Other countries	<u>NA</u>	<u>NA</u>	<u>41,000,000</u>
World total (rounded)	<u>110,000</u>	<u>110,000</u>	<u>130,000,000</u>

RE Oxide basis

Fonte: U.S. Geological Survey, Mineral Commodity Summaries, January 2015

Domain of phosphates, with 3,2% Ce and 2,5% La (close to Indian reserves)

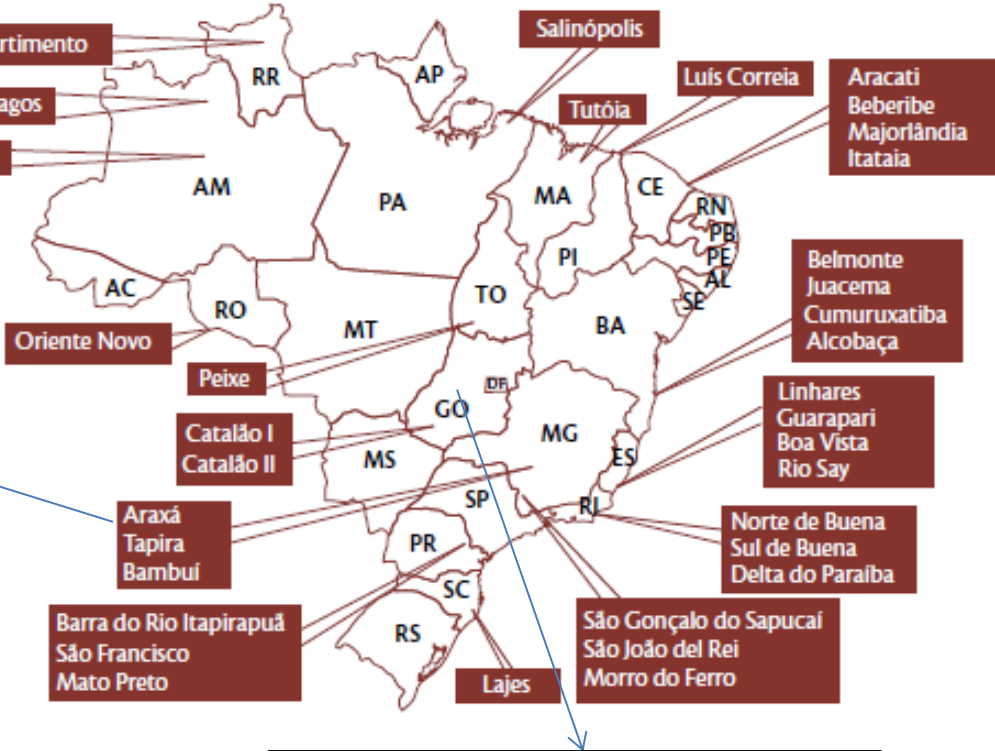
Biological and Indian reserve

Xenotime (1% de RE, mainly Y) – Taboca Mining - Minsur

Tailings of Nb extraction - CBMM

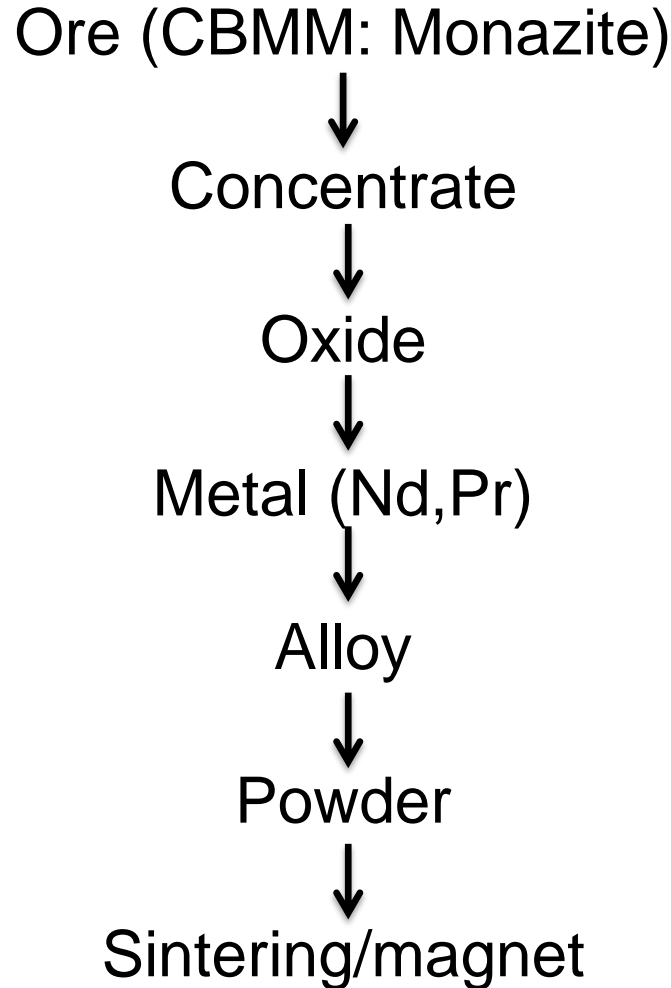
Mineral	%
<b>Bariopyrochlore</b>	<b>4</b>
Limonite, goethite	36
Barite	20
Magnetite	16
Gorceixite	6
<b>Monazite</b>	<b>4</b>
Ilmenite	5
Quartz	4
Others	5
Total	100

Fonte: MCTI, apresentação no CT-Mineral (2010).

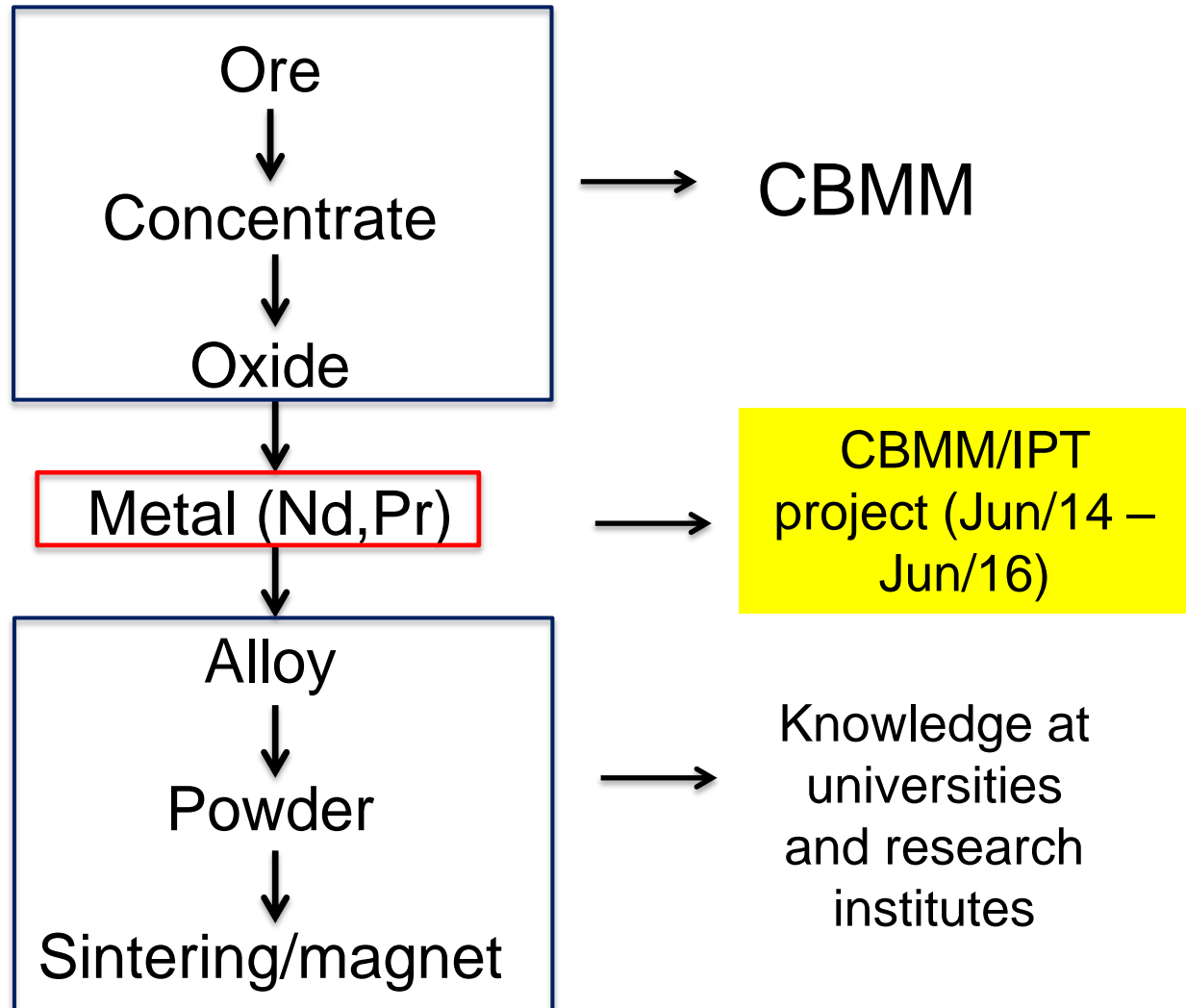


Serra Verde Mining of the Mining Ventures Brazil group (city of Minaçu – GO)

# (Nd,Pr)-Fe-B magnets production



# (Nd,Pr)-Fe-B magnets production



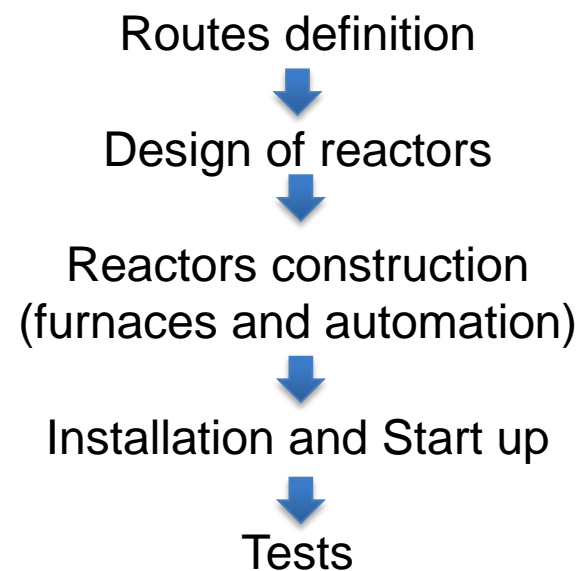
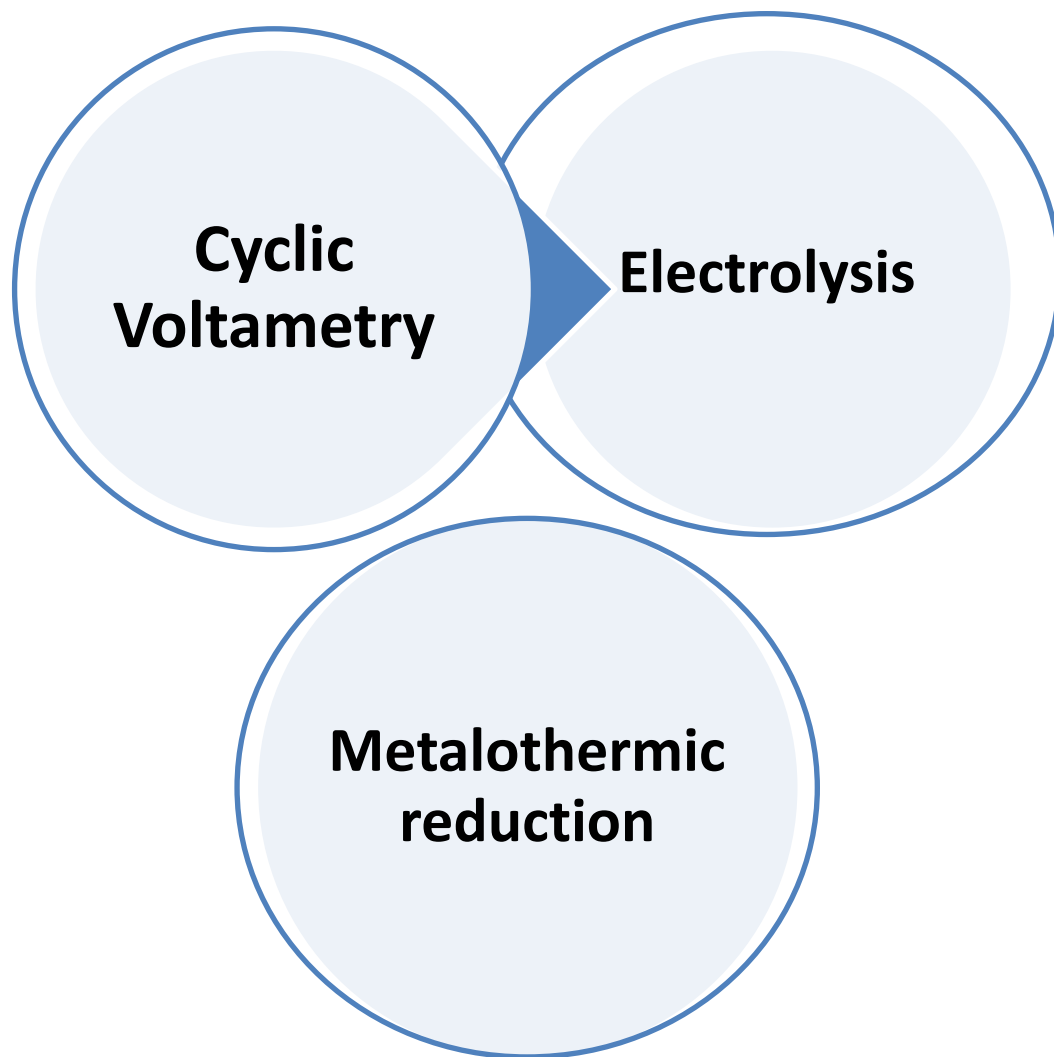
# Routes investigated

## Metalothermic

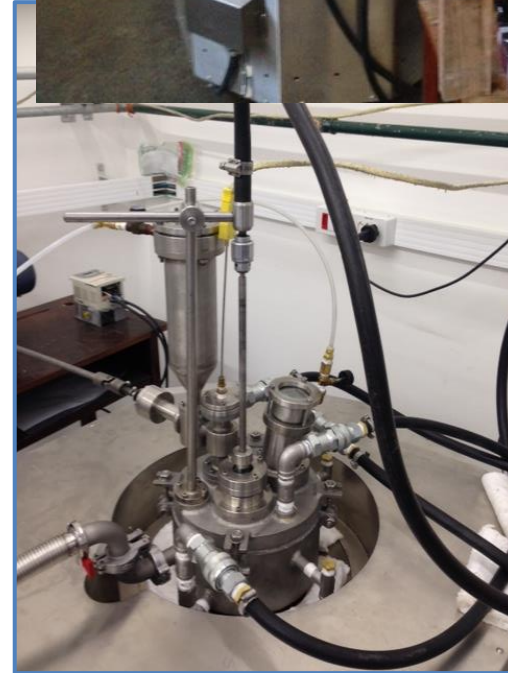
- Production of ingots by batch
- Consumption of reducing agent
- High reactive metals (Ca) – atmosphere controlling
- Salt cannot be recycled (depend on type of salt)

## Electrolysis in molten salt

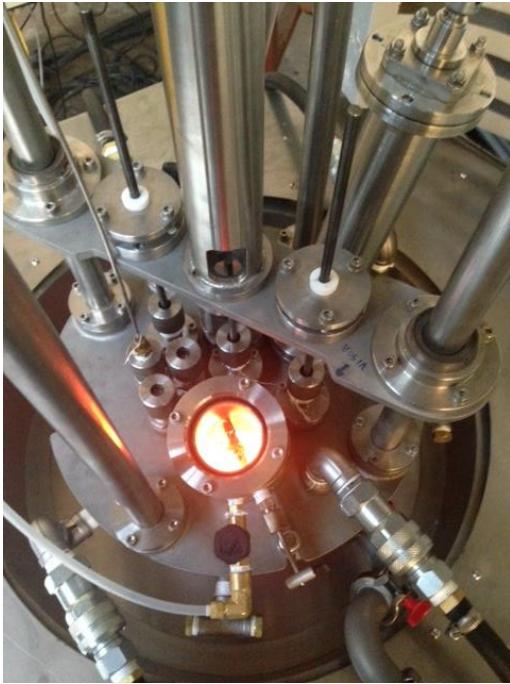
- Continuous process
- Electric energy consumption
- Consumed anodes ( $\text{CO}_2 + \text{CO}_2$  equiv. emission)
- Salt can be recycled (only making up is necessary)



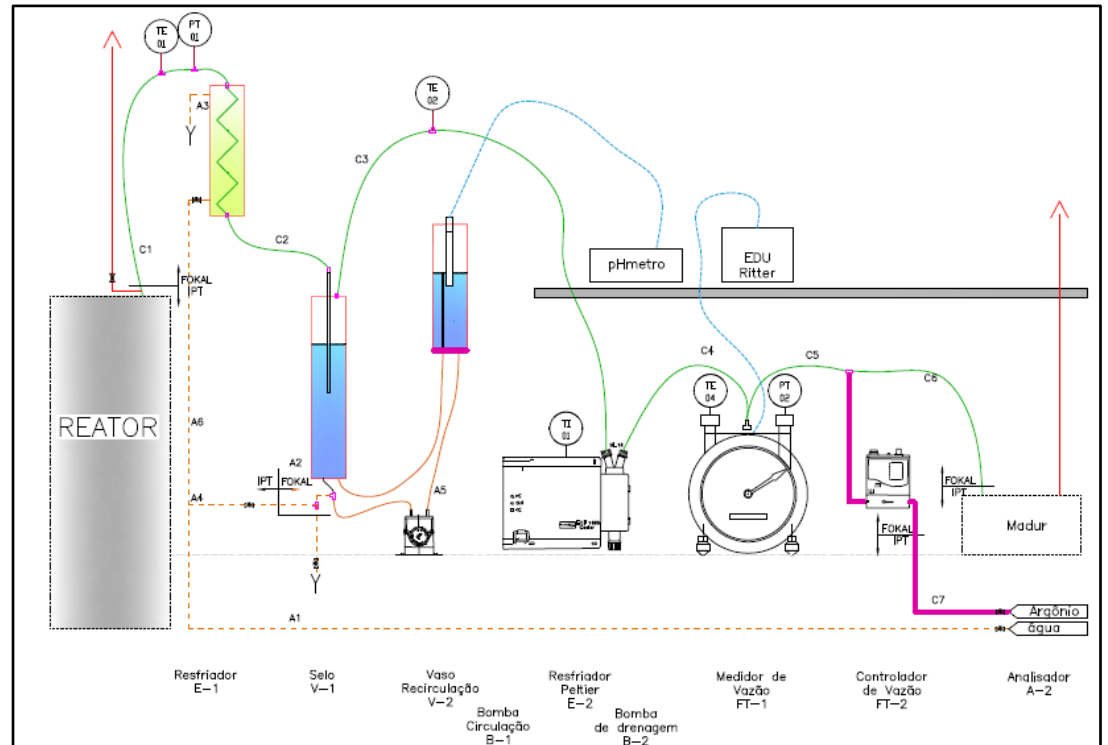
# Cells and reactors



# Electrolysis

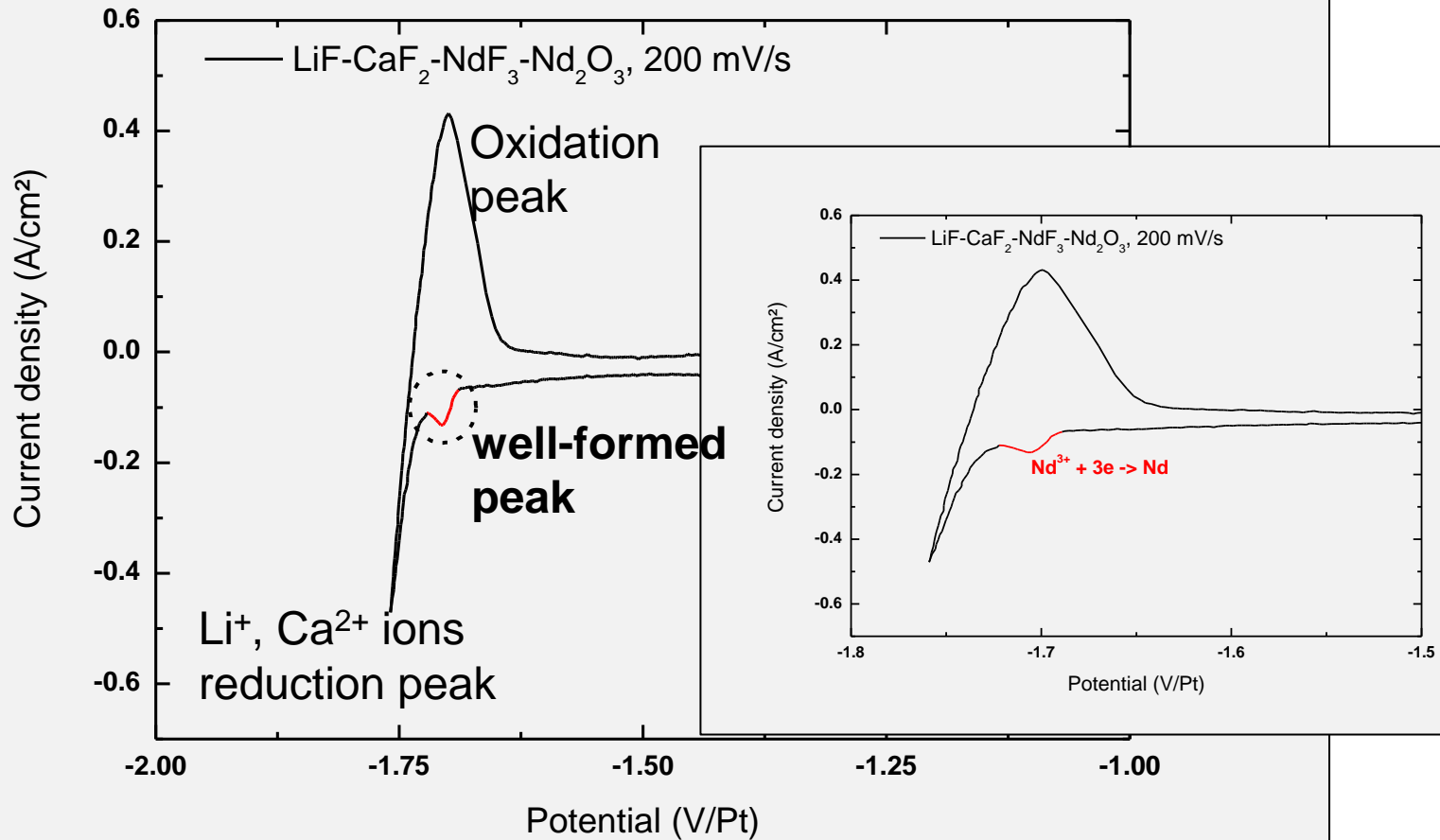


# Cells and reactors



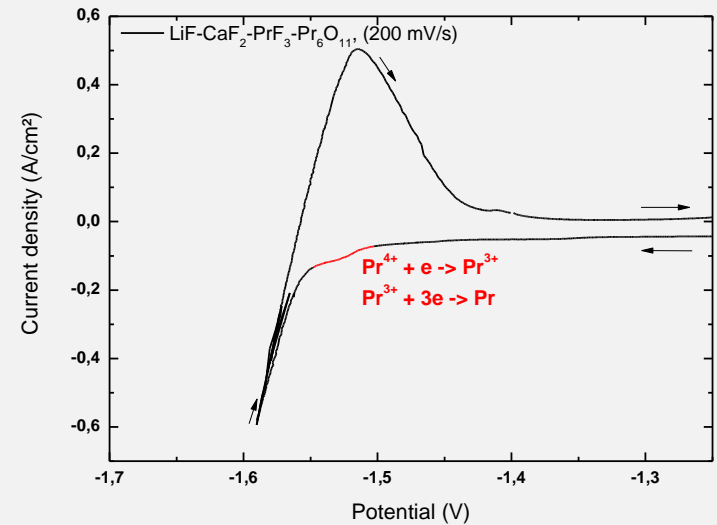
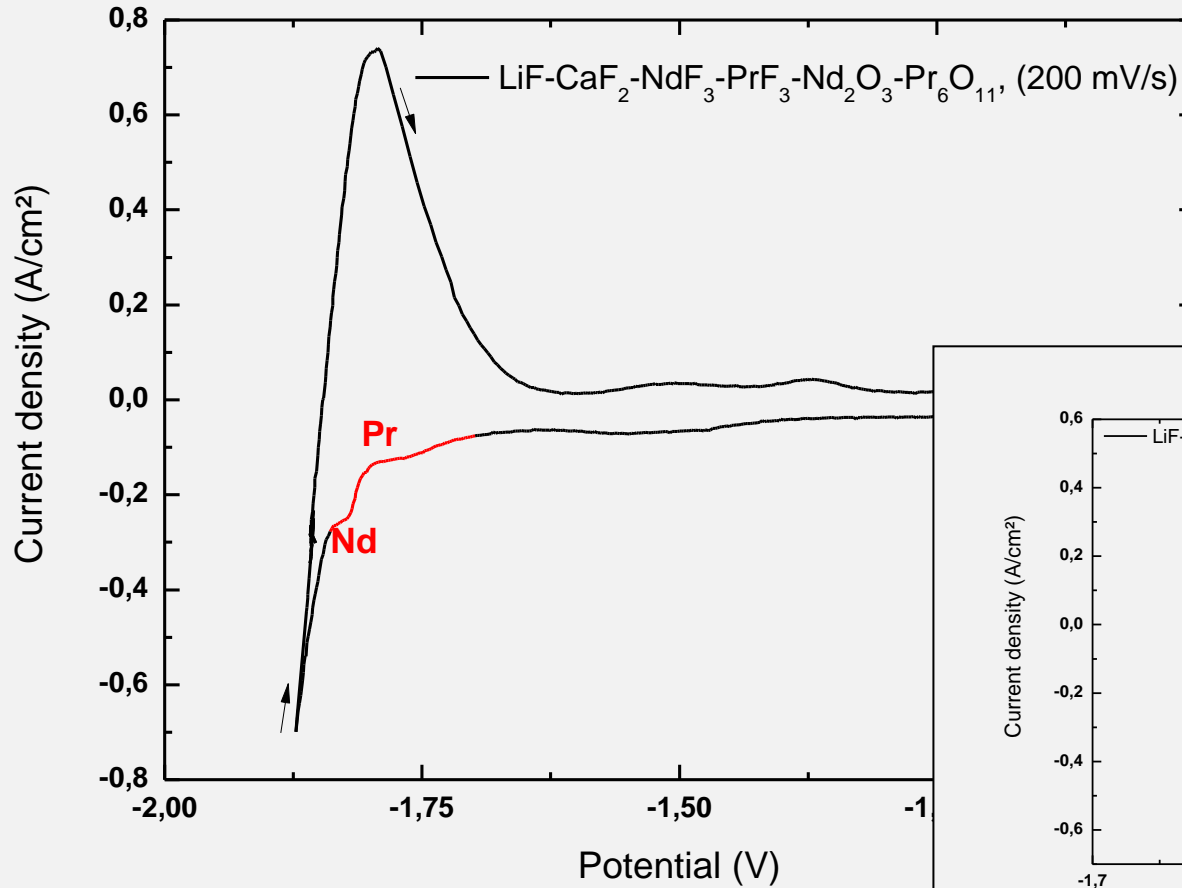
# Cyclic voltammetry experiments

**LiF-CaF<sub>2</sub>-NdF<sub>3</sub>-Nd<sub>2</sub>O<sub>3</sub>**



# Cyclic voltammetry experiments

**LiF-CaF<sub>2</sub>-NdF<sub>3</sub>-PrF<sub>3</sub>-Nd<sub>2</sub>O<sub>3</sub>-Pr<sub>6</sub>O<sub>11</sub>**



# Metalothermic

## - Investigated parameters:

- Temperature
- Time reaction
- Ca excess (solubility of Ca in molten salt)
- Mixing rate
- Ca/salt ratio

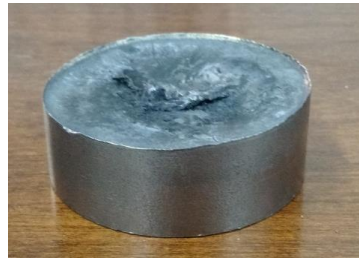
## - Results:

- 200 - 400 g Nd or Di/batch
- Reduction yield: > 95%
- Main impurity: Calcium (Removed by vacuum treatment of liquid didymium)
- Other impurities similar Nd produced with Baotou oxide

# Electrolysis

## - Investigated parameters:

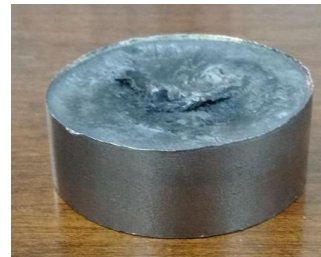
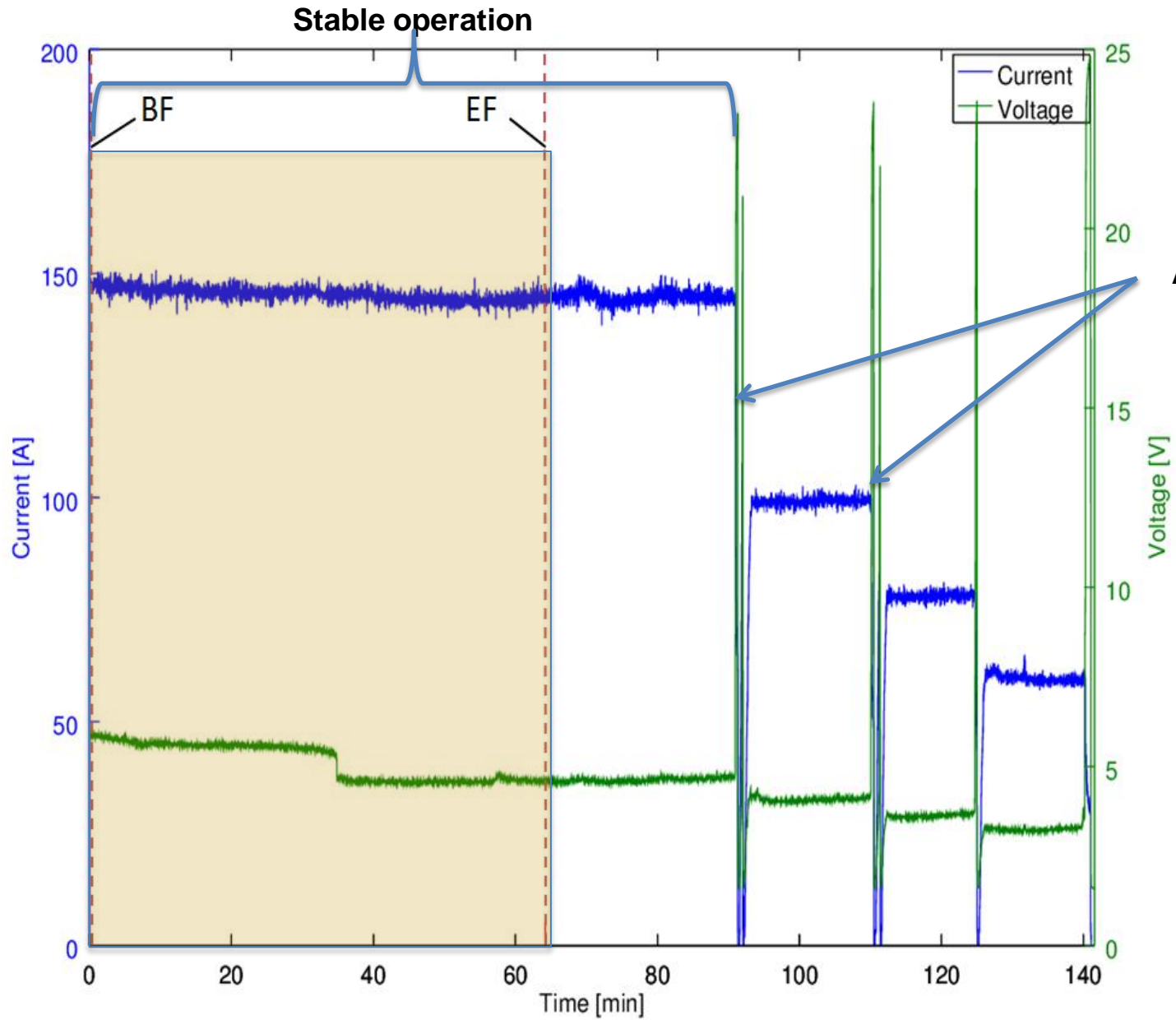
- Salt composition (Fluorides – increase  $\text{Di}_2\text{O}_3$  solubility) – wider operation window
- Oxide feeding rate (control of  $\text{O}^{2-}$  dissolved in molten salt)
- Oxide feeding rate x I (A) – prevent anodic effect (CFx)
- Mixing rate
- Temperature



## - Results:

- 300 - 450 g Nd or Di/batch
- Current efficiency 60/70% ( $\eta = [\text{weight produced Nd/Di}] / [\text{calculated weight Nd/Di production by I (A)}] * 100$ )
- 4-5 Kwh/kg Nd (lab scale)
- Impurities similar Nd produced with Baotou oxide

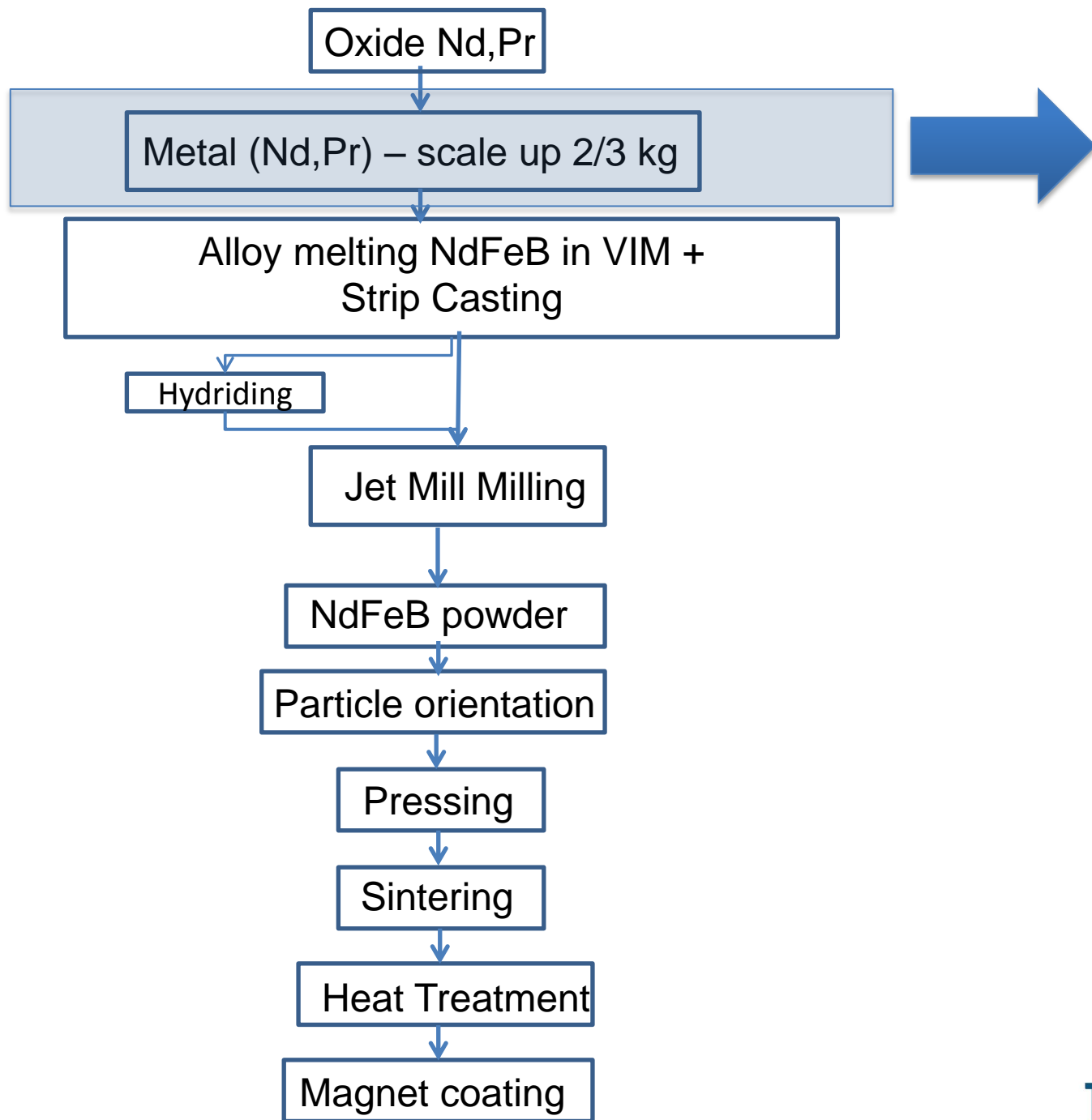
# Electrolysis



Project: Nd/Di production from CBMM oxide

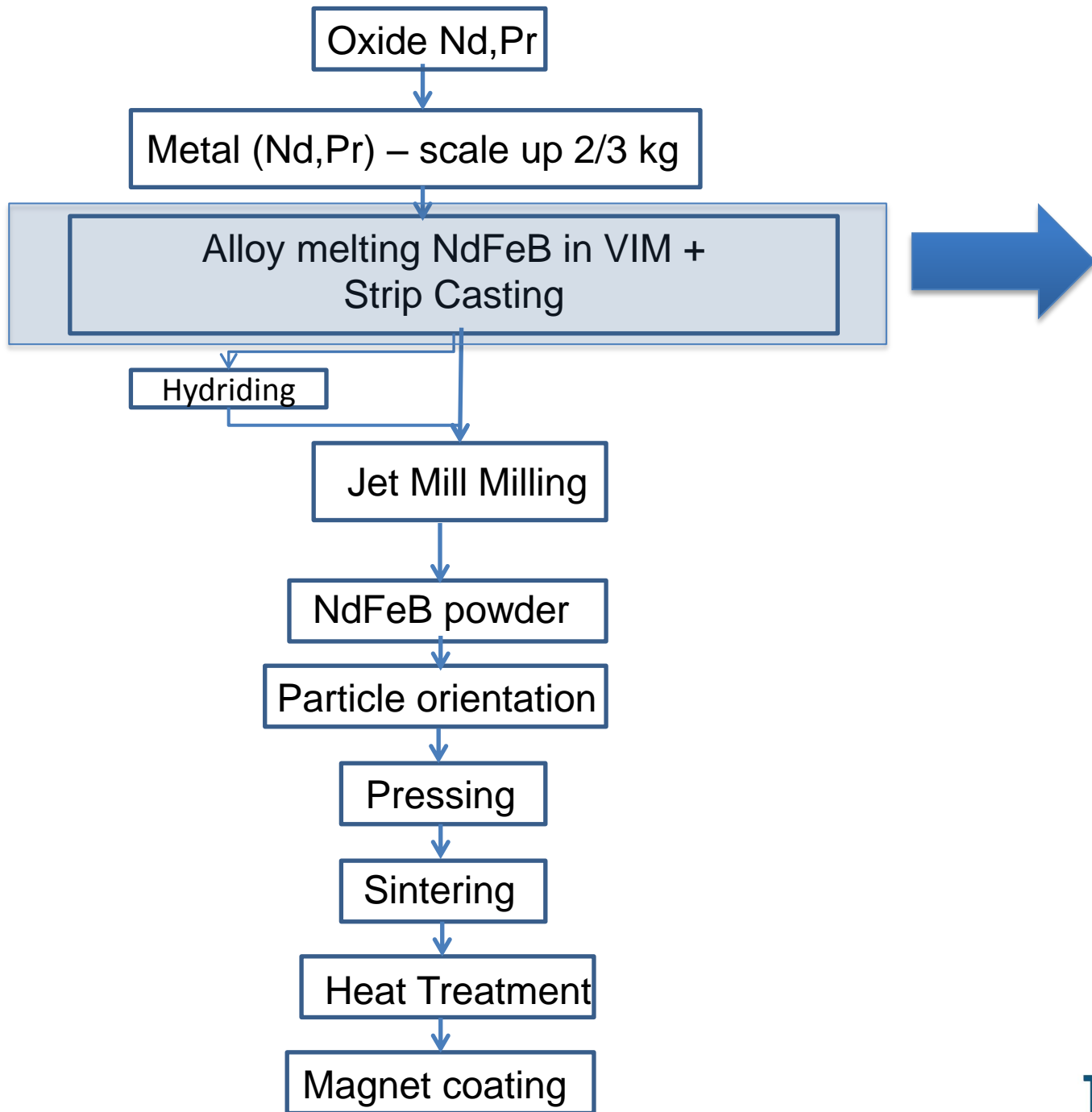


**Next step..... Alloy and magnet  
production**



# Electrolysis – scale up

- Production of 2-3 kg by batch
- Heating source – DC current from electrolysis
- Open cell
- Gas treatment ( $\text{CF}_x$  neutralizing)
- Tapping of liquid Didymium
- Automation for  $\text{Di}_2\text{O}_3$  oxide feeding control (prevent anodic effect)



# VIM

## Main Challenges

Reactivity of Nd/Di with Oxygen

Reactivity of Nd/Di with crucible

Nd/Pr vaporization

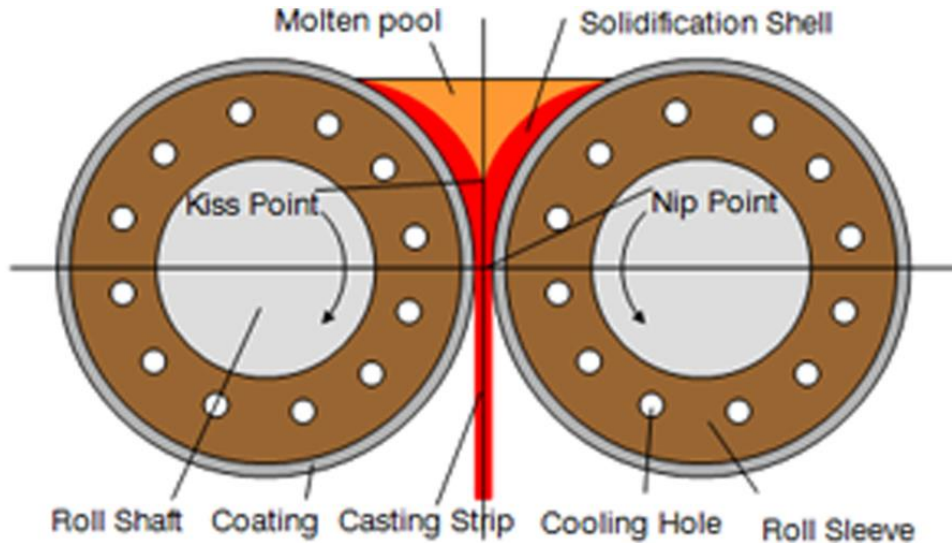


VIM of IPT (capacity: 2-3 kg)

## Raw Materials

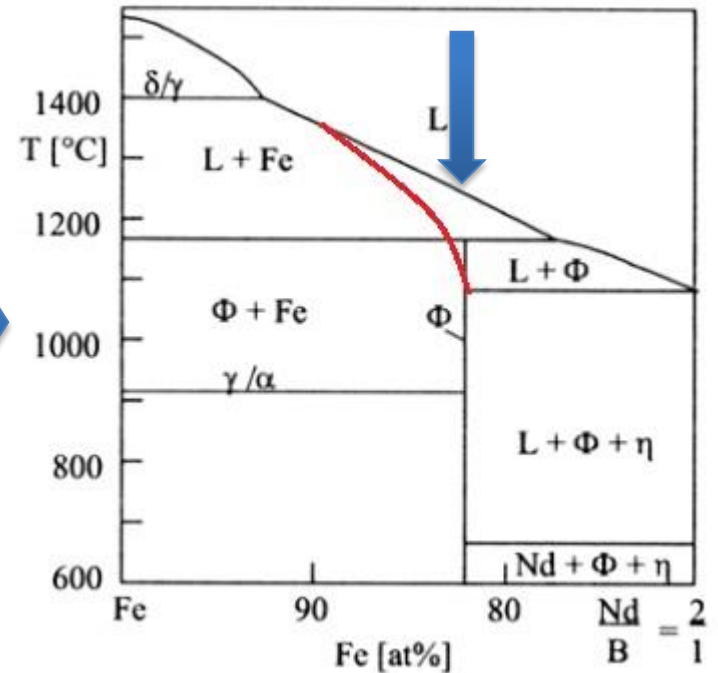
- High purity Fe
- Fe-B or B
- Di

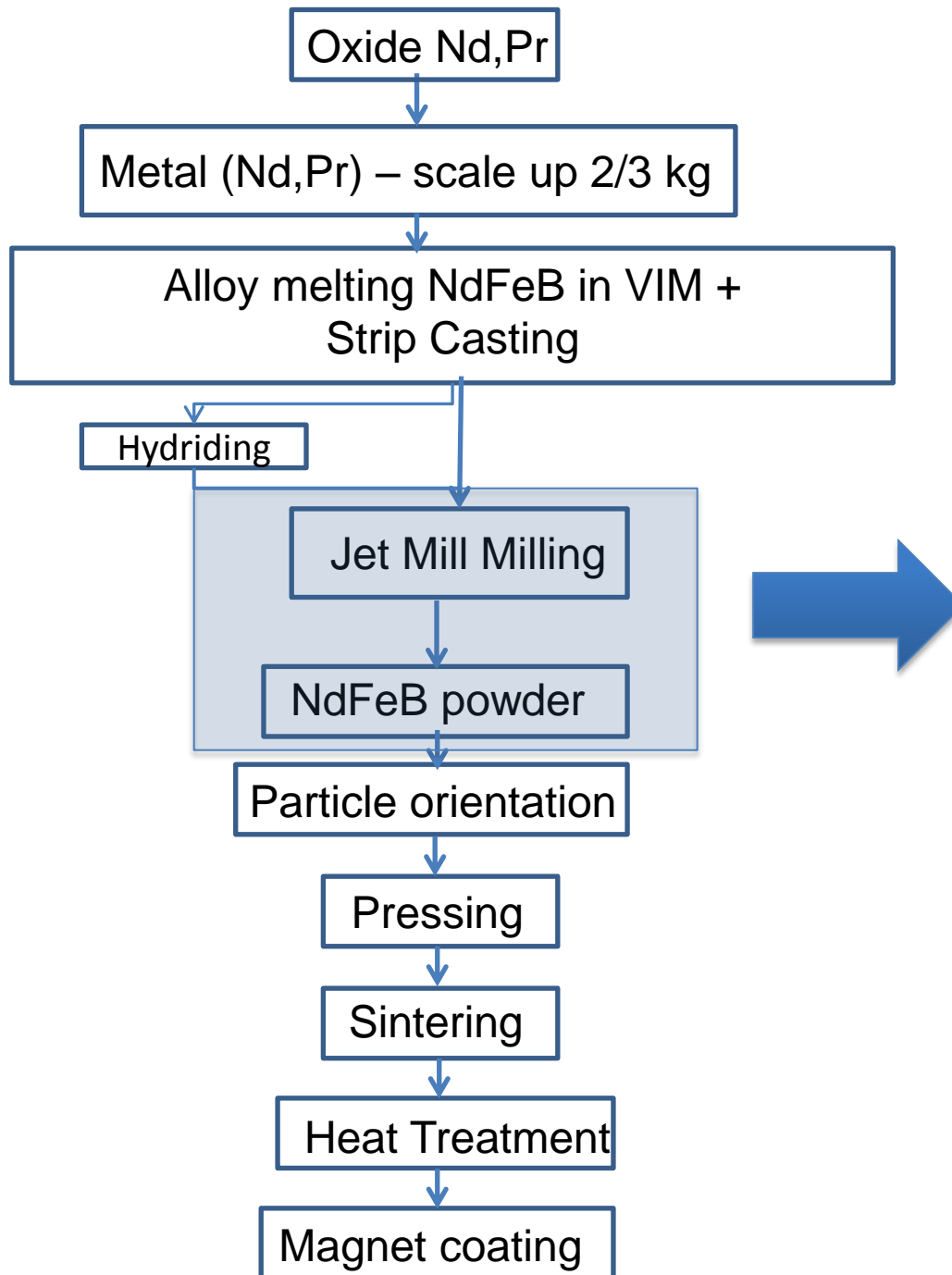
# Fast Cooling – Strip casting



## Main Challenges:

- Prevent Fe( $\alpha$ )
- Nd rich phase in grain boundary ( $\eta$ ) (improving hydriding and magnetic properties)
- Columnar grains - 90% width < 25  $\mu\text{m}$





# Milling – Jet Mill

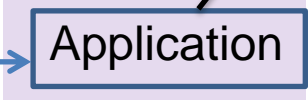
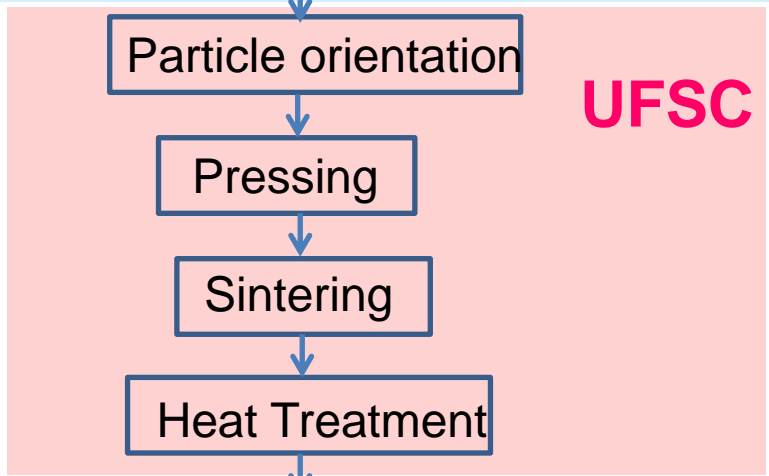
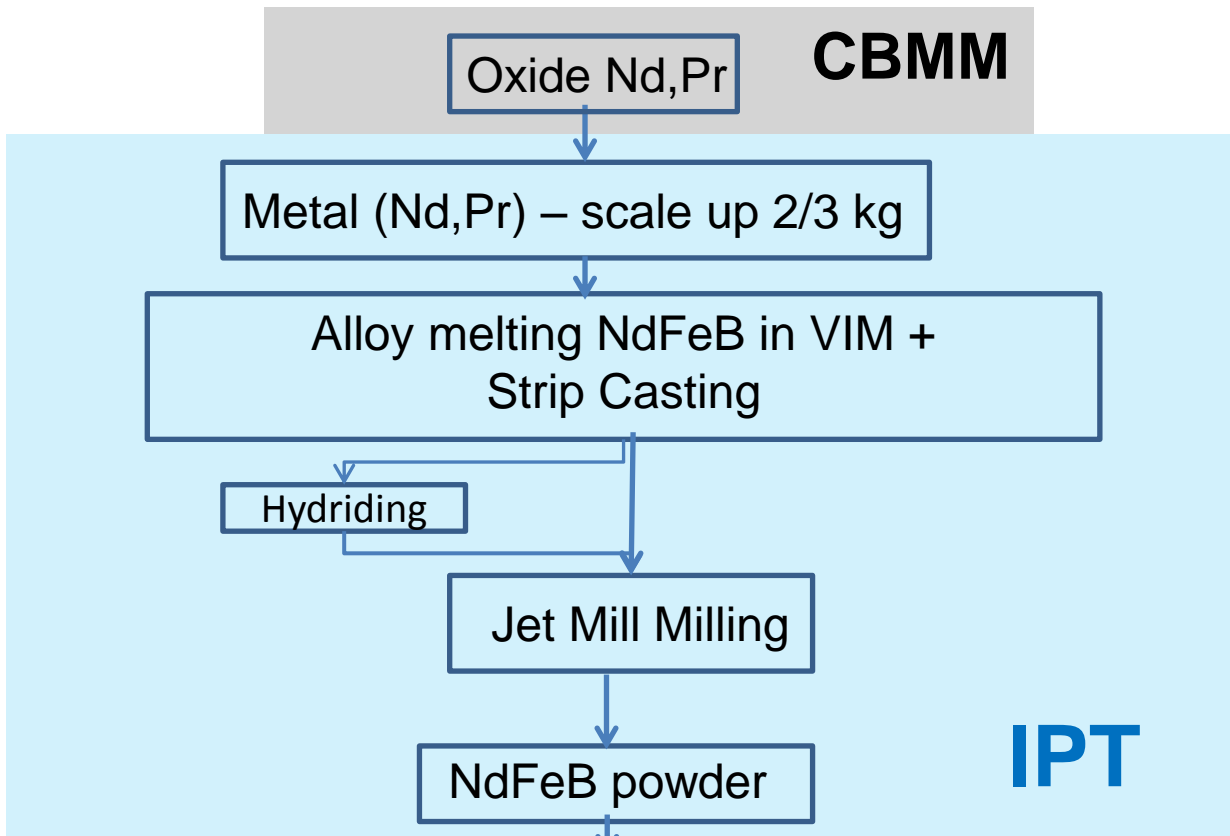
- Pre-milling

- Milling - *Jet Mill*



## Main Challenges:

- Particle size 2-3  $\mu\text{m}$
- O < 1000 ppm



Company interested in magnets application

# Possibilities of cooperation

## Didymium oxide reduction by Electrolysis:

- Identification and quantification of perfluorocarbon gases generated during electrolysis and investigation of routes of gas treatment.
- Determination of didymium (Nd and Pr) oxide solubility and its dissolution kinetic (including a model) in molten fluoride salts (effect of salt chemical composition).
- Possibility of use of inert anode in the didymium/Neodymium electrolysis.
- Fundamental study of kinetic of (Nd,Pr) oxide reduction

# Possibilities of cooperation

(Nd,Pr)-Fe-B alloy and stripcasting:

Effect of cooling rate (wheel speed) of stripcasting and tapping temperature on microstructure of (Nd,Pr)-Fe-B alloys