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Investigation of the susceptibility to preferencial corrosion of pipeline welded joints

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## Investigation of the Susceptibility to Preferential Corrosion of Pipeline Welded Joints

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## What is a welded joint and PWC?









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## Historical background



(1) Rothwell: Neil: Mervyn, 1990 (2) Joosten: Payne, 1988 (3) Olsen: Sundfaer: Enerhaug, 1997 (4) Dawson et al., 1999 (5) Mahajanam: Joosten, 2010 (6) MCintyre: ACHOUR (2016)

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## The problem: gas-exportation pipe corrosion

PWC was observed in gas-exportation in contact with low conductivity CO<sub>2</sub> condensed water





Gas-transportation pipeline



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### Goal

- To find out a criterium for selection of the parent metal and welding consumables for gastransportation pipelines (condensed water and dragged water, which are low conductivity media)
- To establish laboratory facilities to test 79 different circumferential joints with different values
- To choose the best techniques to identify the PWC phenomenon in low conductivity media aiming at understanding the overall processes:
  - o immersion test;
  - o galvanic current measurement;
  - o SVET technique test.





Source: Raman. T. J. Dissertation Universiti Teknologi PETRONAS. 2016.



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## Materials / Parent metals

Seven different parent metals

Parent Metal	Welding process	Most abundant relevant alloy element, in descending order
8" DNVGL SMLS 450	GTAW / GMAW	Si
10" DNVGL SMLS 450	GTAW / GMAW	Cr + Si
12" DNVGL SMLS 450	GTAW / GMAW	Ni + Si + Cu
20" DNVGL SAW 450	GTAW / GMAW / SAW	Ni + Si + Cr + Cu
24" DNVGL SAW 450	GTAW / GMAW / SAW	Si + Cr
8" Forged	GTAW / GMAW	Ni + Cr + Si
20" Forged	GTAW / GMAW / SAW	Ni + Si + Cr + Cu





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## Materials / Welding processes and consumables

<sup>(10)</sup> Most abundant relevant element of the consumable for each welding process

GTAW	GMAW	SAW
↓ Si	↓ Si	Si
↑ Si	↑ Si	Ni + Si
Ni + Si	Ni + Si	Ni + Cr + Si + Cu
Si + Cr + Ni	Ni + Si + Cu	-X-
Ni + Si + Cu + Cr	Si + Cu + Cr + Ni	-X-

#### **Total of 79 welded joints**





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## Materials / Specimens

- Immersion-test samples:
- Girth welds =
  SAW (wider)/ GTAW / GMAW (narrower)



#### Sample ready to be tested



470 Scotch tape to protect the surface This area was used as a reference for thickness-loss measurement

#### Sample after the test



Non-corroded area (reference)



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## Materials / Specimens

#### **GCM TEST**











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## Preliminary tests

- Several preliminary tests were conducted for the selection of methodologies for low conductivity media.
- Galvanic current measurement: anodic and cathodic regions of the galvanic couple MS/MB+ZTA alternate over time.
   No clear indication of PWC occurrence.

<sup>(10)</sup> Immersion test and SVET technique using the entire joint were selected.









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#### Immersion tests 48 h (condensed water) 120 h (dragged water)

TE





## Identification of PWC occurrence by confocal microscope



Light HAZ preferential attack close to HAZ/WM fusion line: the galvanic action is restricted to the distance less than 100  $\mu$ m from fused line (FL) because of the low conductivity of the condensed water (45  $\mu$ S/cm)





## Identification of PWC occurrence by confocal microscope



**Light HAZ preferential attack close to HAZ/WM fusion line**: the galvanic action is restricted to the distance less than 100  $\mu$ m from fused line (FL) for the condensed water (**45**  $\mu$ **S**/**cm**) and less than 200  $\mu$ m for the dragged water (**190**  $\mu$ **S**/**cm**).





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## SVET Results

- Condensed water WJ1
- <sup>(10)</sup> WM is the cathode in the vicinity of the fused line (in agreement with the immersion tests)







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## Statistical analyses

Tor the condensed water:

The ANOVA and regression analyses indicated that only the maximum corrosion rates (MCR) and maximum gouge heights (MGH) were affected by the composition of the WM or PM;

There is no dependence of the corrosion rates and the welding processes.





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## Statistical analyses

#### **•** For the condensed water:

The equation obtained for the maximum groove heights (MGH) was:

 $MGH = 1.71 - 0.58(Ni_{PM} - Ni_{WM}) - 1.49(Cu_{PM} - Cu_{WM}) - 1.36(Cr_{PM} - Cr_{WM})$ 

With correlation coefficient of 84 % and considering 22 welded joints excluded from the regression

PM and WM difference	Chemical composition valid range
ΔΝί	-2.172 a 0.243
ΔCu	-0.517 a 0.127
ΔCr	-0.443 a 0.442





## **Overall results**

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#### For condensed water

 $\Delta < 0$  : WM is cathode  $\Delta > 0$ : PM is cathode

#### $\Delta$ > 0.3 high probability for WM/HAZ corrosion



The negative delta-parameter criterium, within the values considered in this study (-2.53 $\leq$  $\Delta$ <0), is capable of reliably predicting the cathodic nature of the WM of a welded joint

## **Overall results**

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The positive values considered within the test matrix of this study (0< $\Delta$ ≤0.95) are not able to predict precisely whether preferential corrosion will occur in the WM/HAZ or not, and the division into two blocks, before and after the value of 0.3, is not valid

## **Overall results**

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The dragged water is a favorable medium for the formation of a protective layer of iron carbonate, but the test duration did not allow the layer to consolidate.



HAZ

WM



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• For low conductivity media with  $CO_2$  (condensed water):

 the △ parameter accurately indicates only the cathodic behavior of the weld metal.



Conclusions



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## Conclusion

- $^{\odot}$  The immersion tests showed the preferential corrosion was restricted in the fusion line, reaching 100  $\mu m$  to the condensed water and 200  $\mu m$  to the dragged water
- <sup>∞</sup> For the condensed water, the  $\triangle$  parameter was valid only for -2.53≤ $\Delta$ <0, which indicates the cathodic behavior of the WM
  - For positive values,  $0 < \Delta \le 0.95$ , it was not observed the cathodic behavior of the parent metal (or HAZ) or the occurrence of PWV in the WM/HAZ region, above or below 0.3
- Por the dragged water, 120 h of immersion was not enough to consolidate the iron carbonate layer
- <sup>(D)</sup> Statistical analysis was useful for identifying the parameters influencing corrosion









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# Thank you for your attention!

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