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Oxygen consumption in confined environments: seawater in contact with stainless steels

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Zehbour Panossian - IPT





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Oxygen consumption in confined environments in seawater in contact with stainless steels





Motivation

- ✓ Hibernation (submersing in seawater and filling with seawater) of flexible pipes and umbilicals occurs when they remain inactive for long periods.
- ✓ In this conditions, a special confined environment is established: low volume of seawater (V) and the large area of metallic surface (A).
- ✓ Under these conditions, rapid oxygen consumption occurs.
- ✓ There are doubts about the reasons for the rapid oxygen consumption:
 - some authors attribute this consumption to the formation of the passive layer over a large surface area of the metal alloys used (stainless steels AISI 304, AISI 316, and duplex steel SAF 2205).



Objective

Understanding the causes for the rapid consumption of O_2 during hibernation with natural seawater, focusing on the metal materials normally used in flexible pipes, and considering the variations of:

- ✓ **material:** AISI 304, AISI 316, and duplex steel SAF 2205;
- ✓ **temperature:** 5 °C, 15 °C, and 25 °C;
- ✓ **medium:** natural and synthetic seawater (sterilized and non-sterilized).

Test Matrix



Steels	Natural seawater			Sterilized synthetic seawater			Sterilized natural seawater		
AISI 304	5 °C	5 °C	5 °C	5 °C	5 °C	5 °C	5 °C	5 °C	5 °C
	15 °C	15 °C	15 °C	15 °C	15 °C	15 °C	15 °C	15 °C	15 °C
	25 °C	25 °C	25 °C	25 °C	25 °C	25 °C	25 °C	25 °C	25 °C
AISI 316	5 °C	5 °C	5 °C	5 °C	5 °C	5 °C	5 °C	5 °C	5 °C
	15 °C	15 °C	15 °C	15 °C	15 °C	15 °C	15 °C	15 °C	15 °C
	25 °C	25 °C	25 °C	25 °C	25 °C	25 °C	25 °C	25 °C	25 °C
SAF 2205	5 °C	5 °C	5 °C	5 °C	5 °C	5 °C	5 °C	5 °C	5 °C
	15 °C	15 °C	15 °C	15 °C	15 °C	15 °C	15 °C	15 °C	15 °C
	25 °C	25 °C	25 °C	25 °C	25 °C	25 °C	25 °C	25 °C	25 °C

Total possible tests: 81

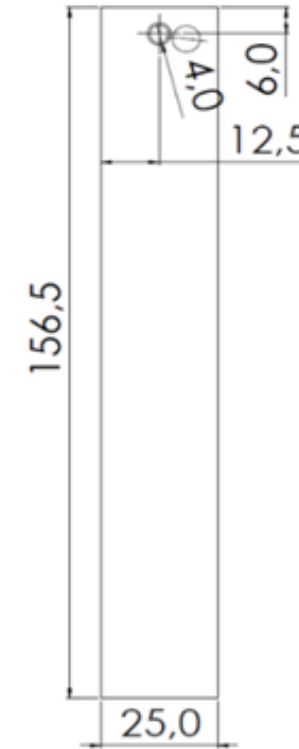
Selected tests: 34

Methodology

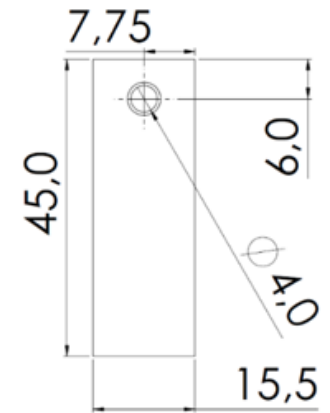


The total area of coupons was fixed to achieve a V/A of 1.7 mL/cm²:

- ✓ five coupons with dimensions 156.5 mm x 25 mm x 1.4 mm;
- ✓ two biocoupons with dimensions 45 mm x 15.5 mm x 1.4 mm;
- ✓ surface finishing:
 - as received;
 - special treatment:
 - immersion in 10 % nitric acid at 60 °C;
 - washing with water;
 - immersing in 70 % ethanol;
 - dried (inside a microbiological hood);
 - immersed in the test vessel inside a microbiological hood.



Cupom



Biocupom



Test environment

Four types of test medium were used:

- ✓ **natural seawater** collected from the São Sebastião canal, transported on the same day of collection, filtered with a 50 μm filter, and preserved at 5 °C until the start of the test, which occurred within 48 h after collection;
- ✓ **natural seawater** (the same as cited above) **sterilized**;
- ✓ **sterilized synthetic seawater** according ASTM D1141, without heavy metals;

Sterilization: autoclave at 121 °C \pm 5 °C under 1.1 atm for 15 min

Determination of the O₂ content



Adhesive optical sensors (PSt3 - Presens Precision Sensing):

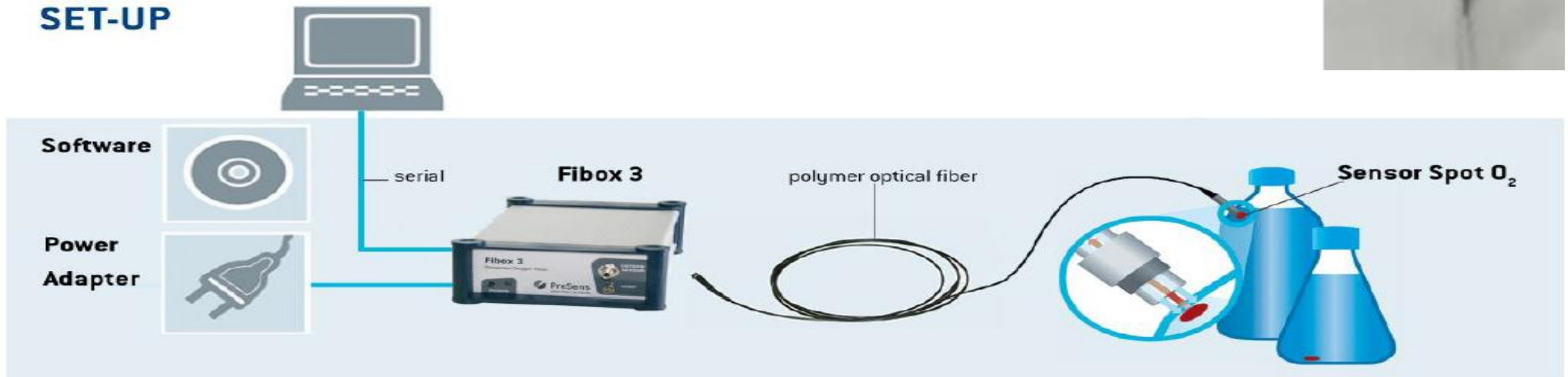
- ✓ sensors glued to the inner wall of the test vessel;
- ✓ a small glass tube was glued to the external wall in front of the sensor to insert the reading sensor.



Black discs are adhesive optical sensors!
Optical luminescence O₂ sensors



SET-UP

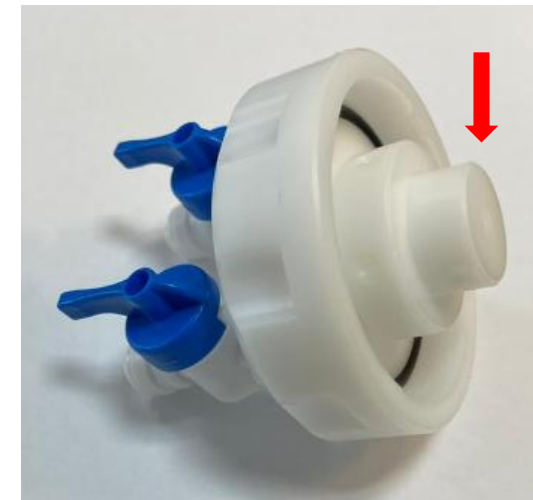
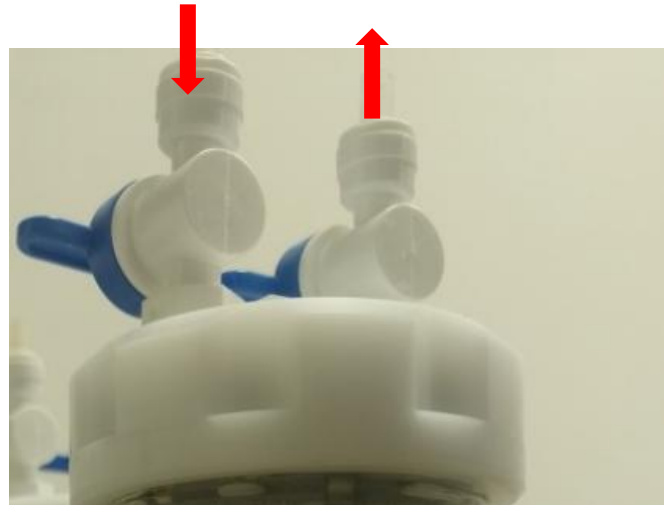


Preparation and sterilization of the test devices



The test vessel and its lid were specially prepared:

- ✓ threaded glass cylinder 16 cm high and 8 cm in diameter;
- ✓ a Teflon lid:
 - at the top side, provided with one gas inlet and one gas outlet (N_2 or O_2);
 - at bottom, adapted for a device for installing test specimens.



Sterilization by rinsing with 70 % ethanol or exposing to UV light.

Assembling



Steps:

- installation of the specimens and biocoupons in the lid and closing of the test vessel;
- transfer of test medium to test vessel;
- adjusting the desired temperature;
- bubbling with pure hospital-grade O₂ gas until reaching a content just above 8 mg/L;
- Monitoring the O₂ content for 60 days.



Monitoring the Temperature and the O₂



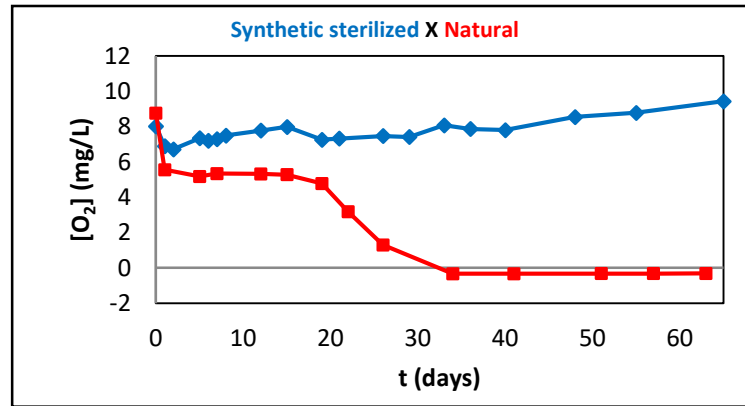
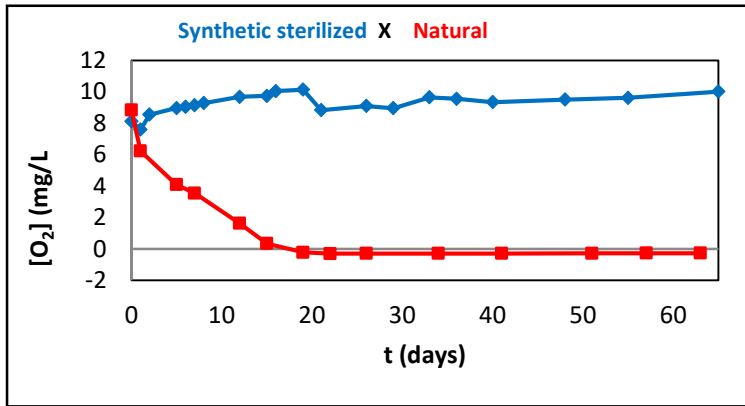
5 °C e 15 °C



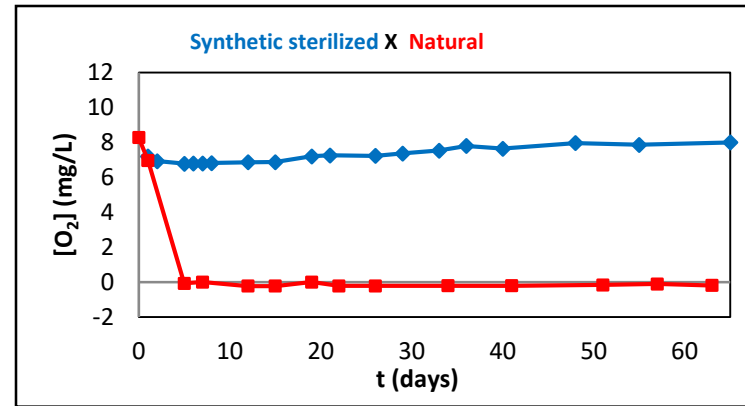
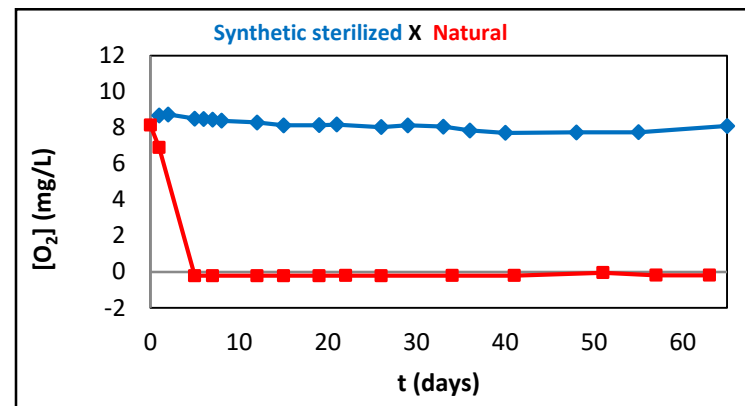
25 °C

AISI 304

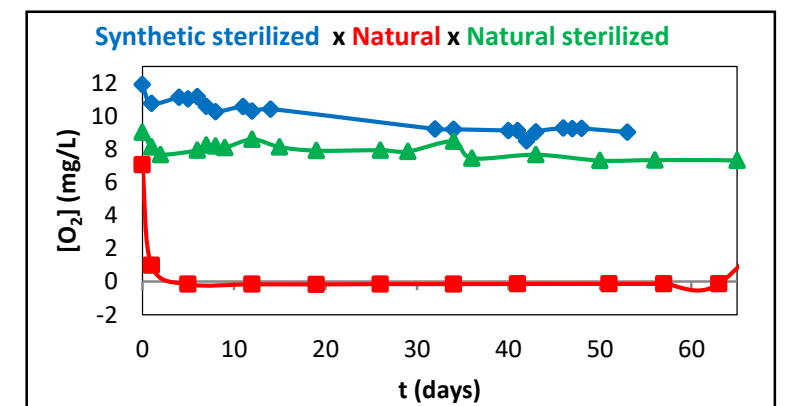
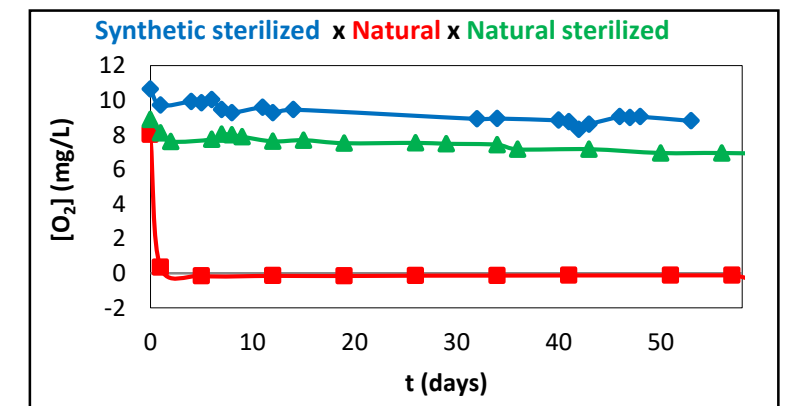
5 °C



15 °C

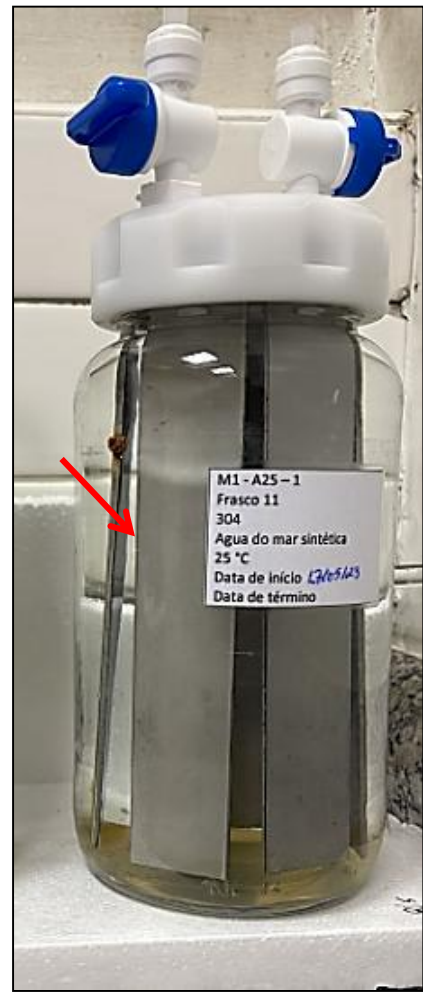


25 °C



- **Rapid consumption of O_2 in natural seawater:** the faster the consumption, the higher the temperature.
- ◆ **No O_2 consumption in synthetic sterilized seawater** at 5 °C and 15 °C
- ◆ ▲ **Slight O_2 drop in sterilized seawater** at 25 °C

Synthetic sterilized 25 °C



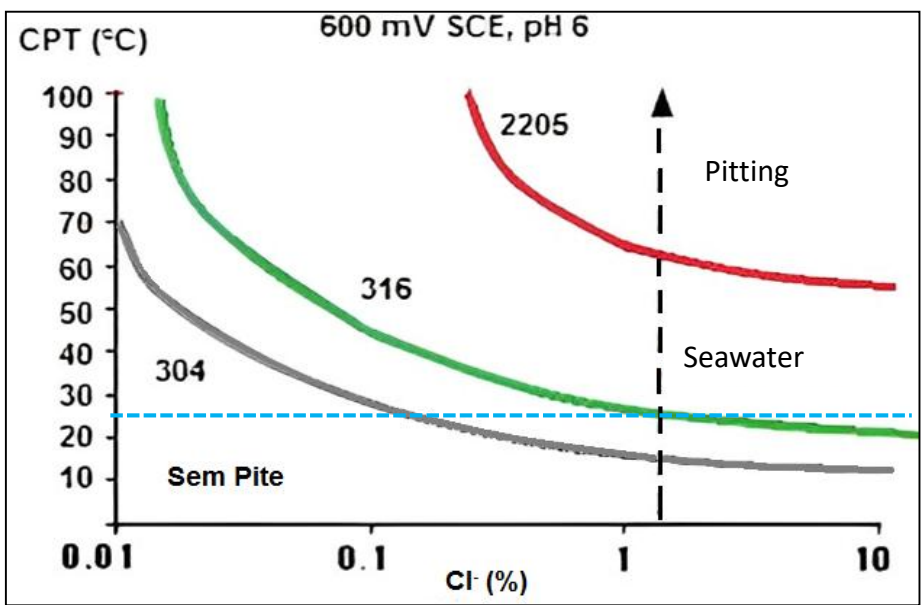
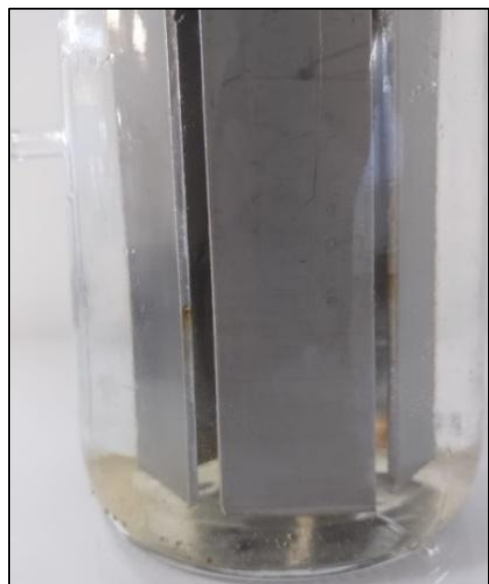
Natural sterilized 25 °C



Synthetic sterilized 15 °C

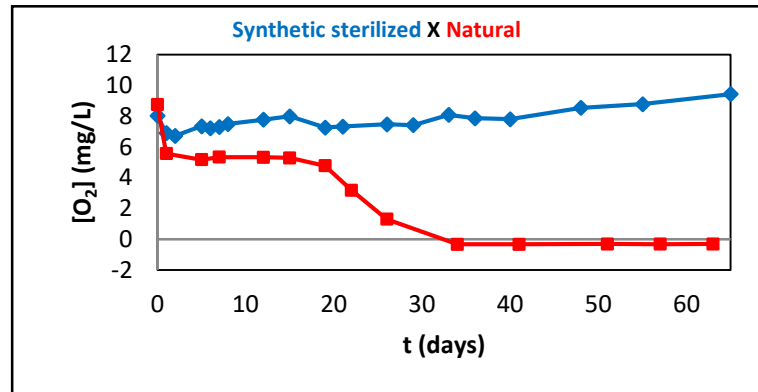
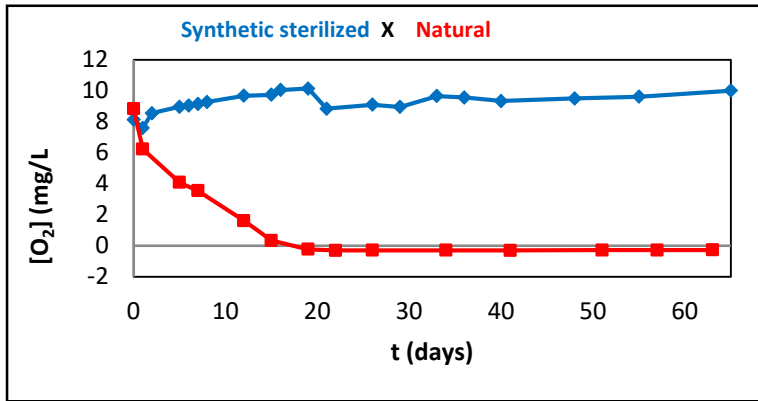


Natural sterilized 25 °C

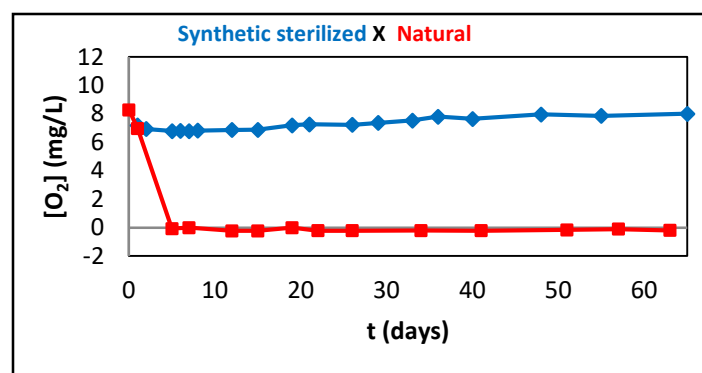
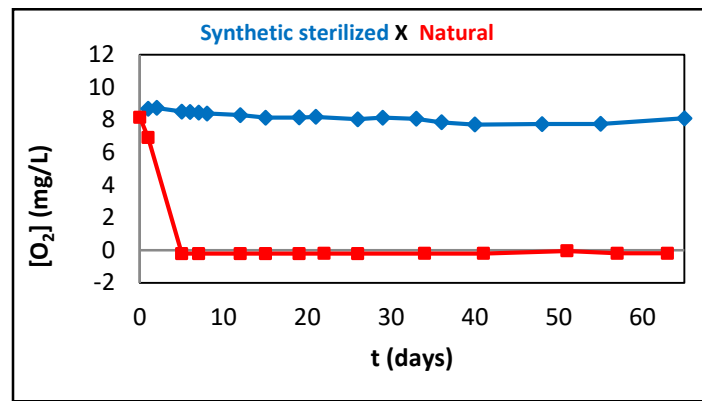


AISI 304

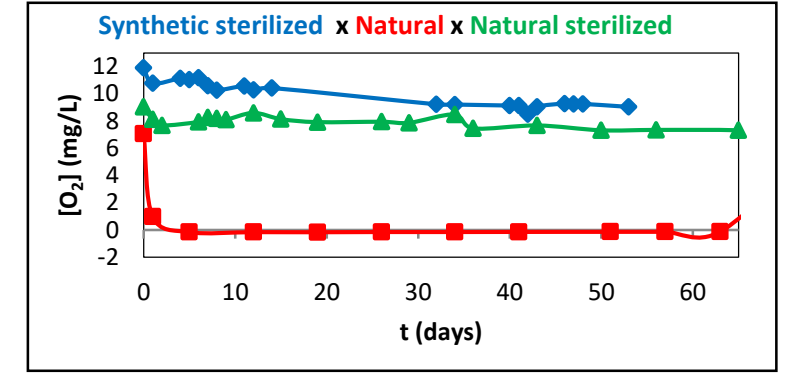
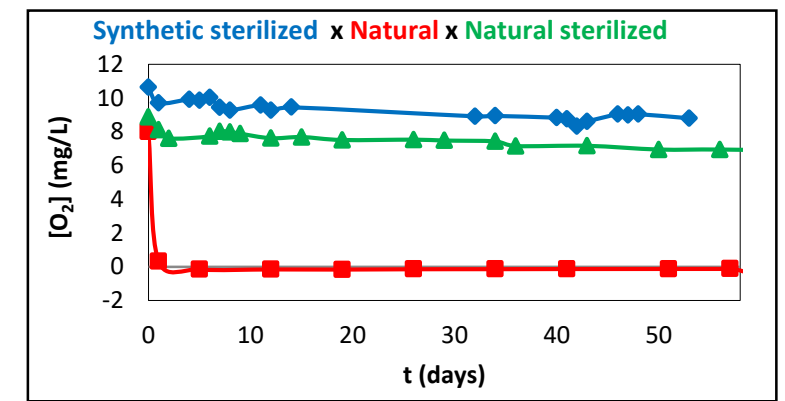
5 °C



15 °C



25 °C

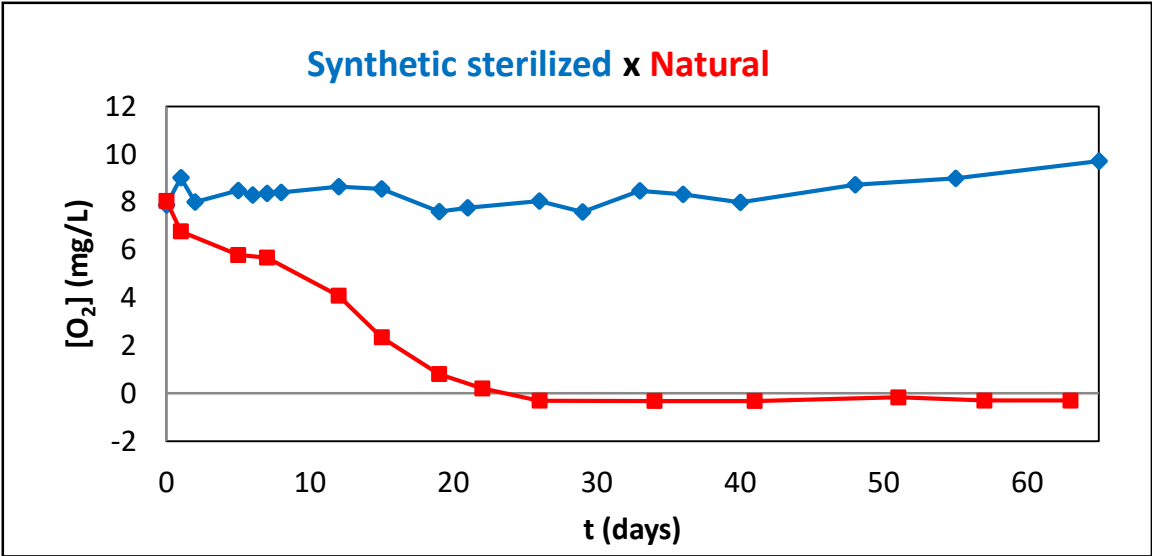
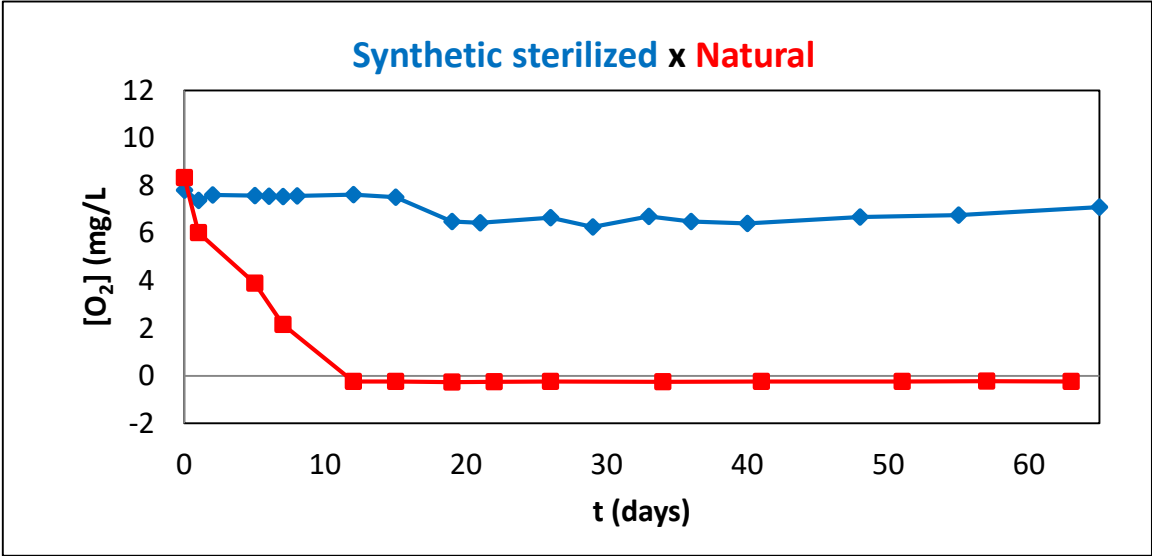


- **Rapid consumption of O_2 in natural seawater:** the faster the consumption, the higher the temperature.
- ◆ **No O_2 consumption in synthetic sterilized seawater** at 5 °C and 15 °C
- ◆ ▲ **Slight O_2 drop in sterilized seawater** at 25 °C

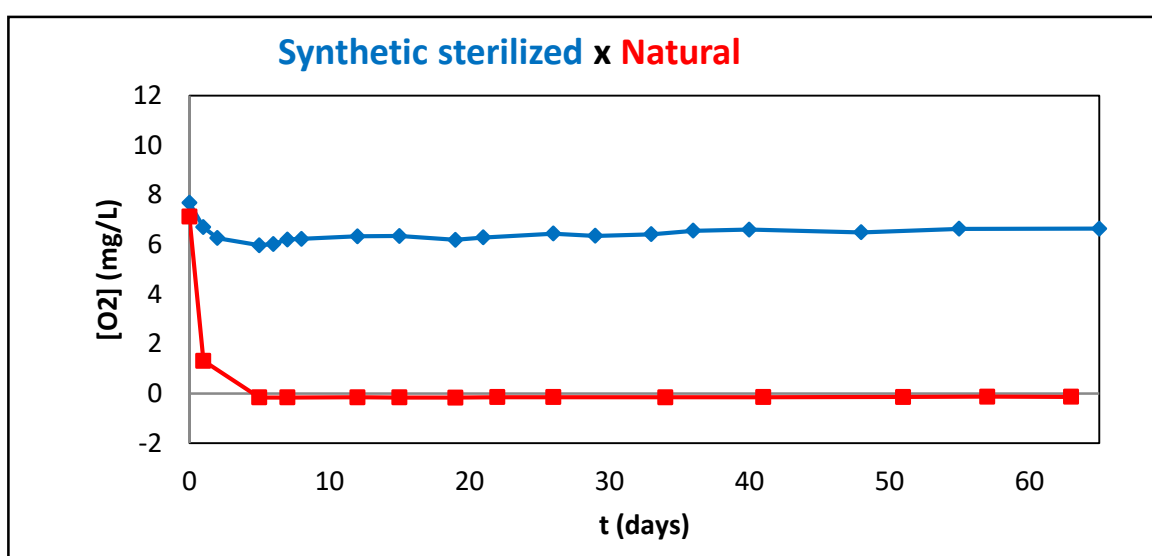
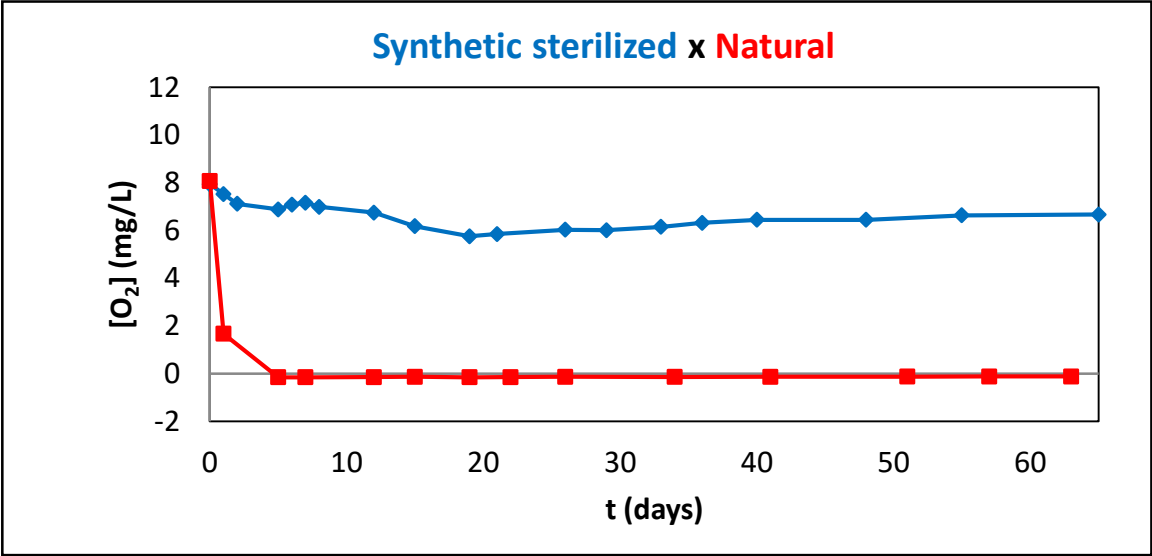
Conclusion

O_2 consumption is associated with the presence of microorganisms present in non-sterilized seawater and not with metal passivation as mentioned in the literature.

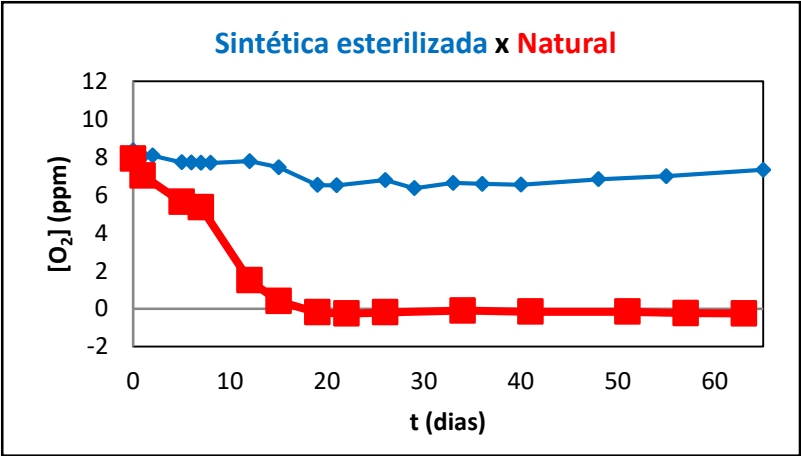
5 °C



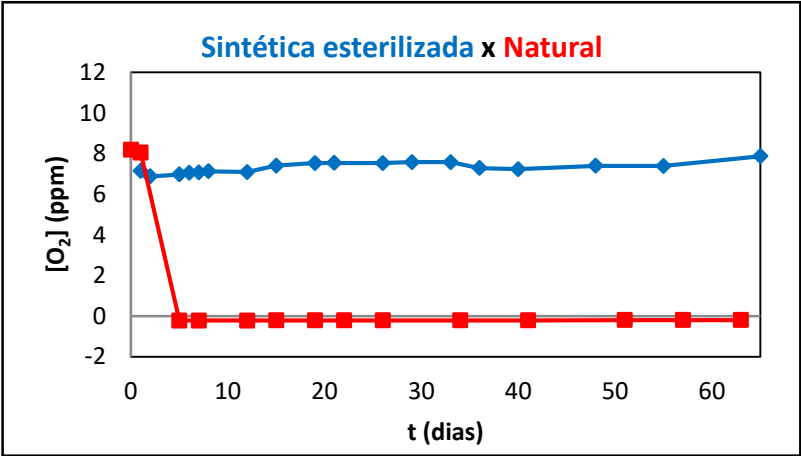
25 °C



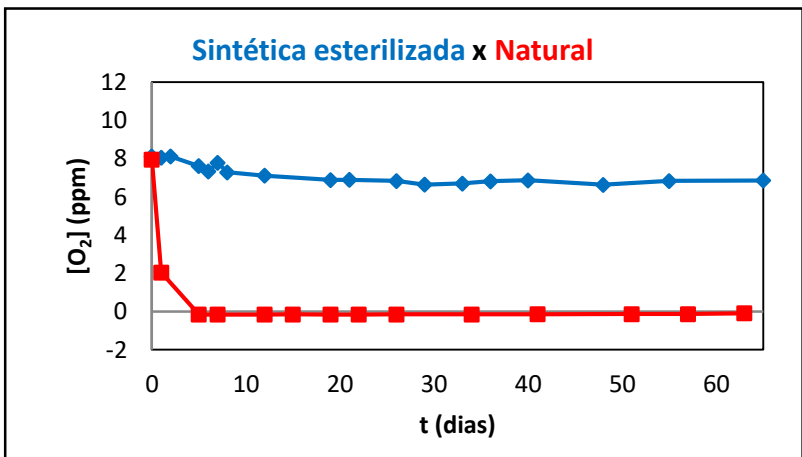
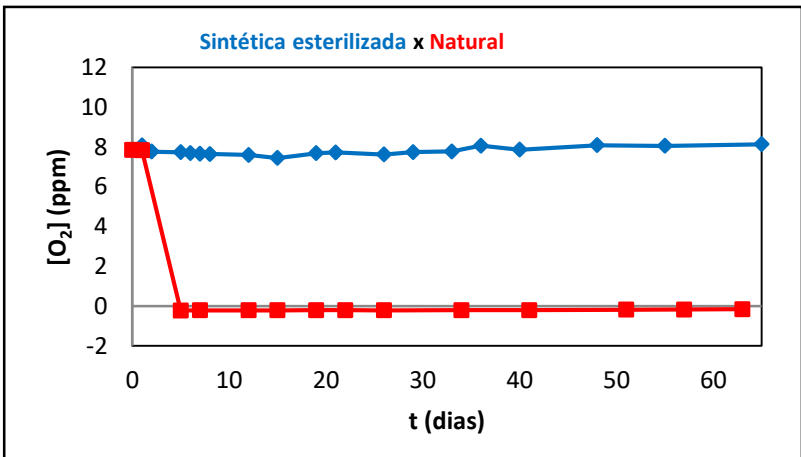
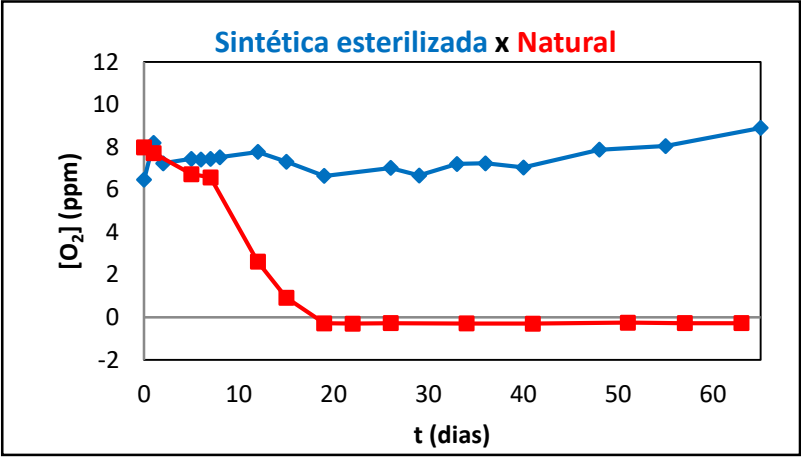
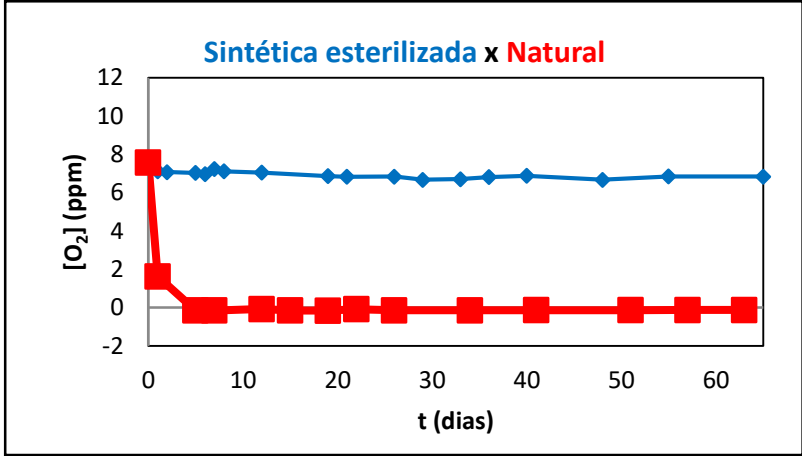
5 °C



15 °C



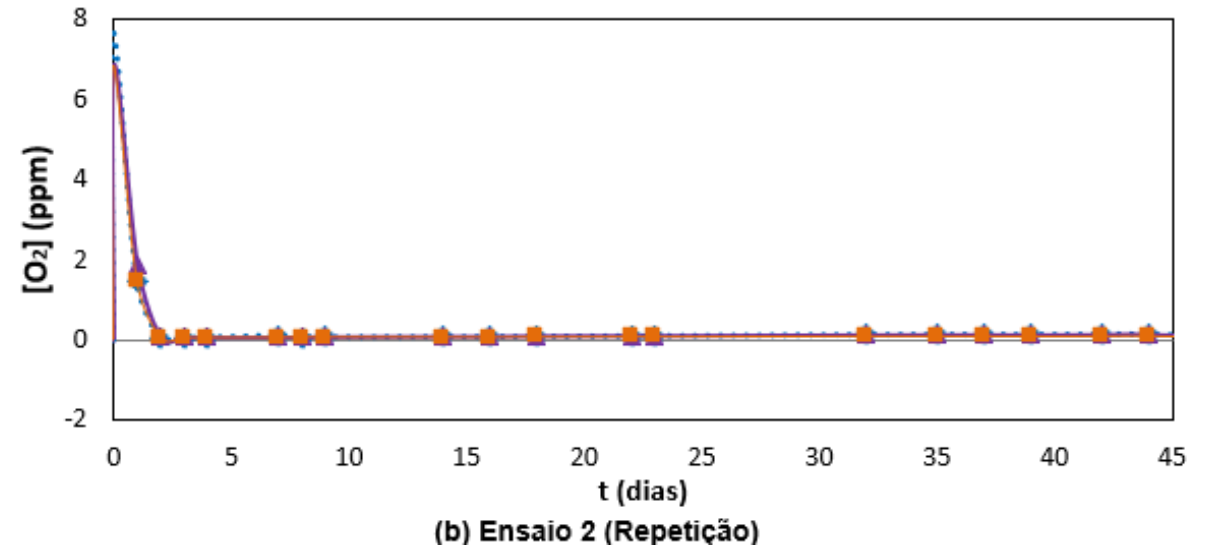
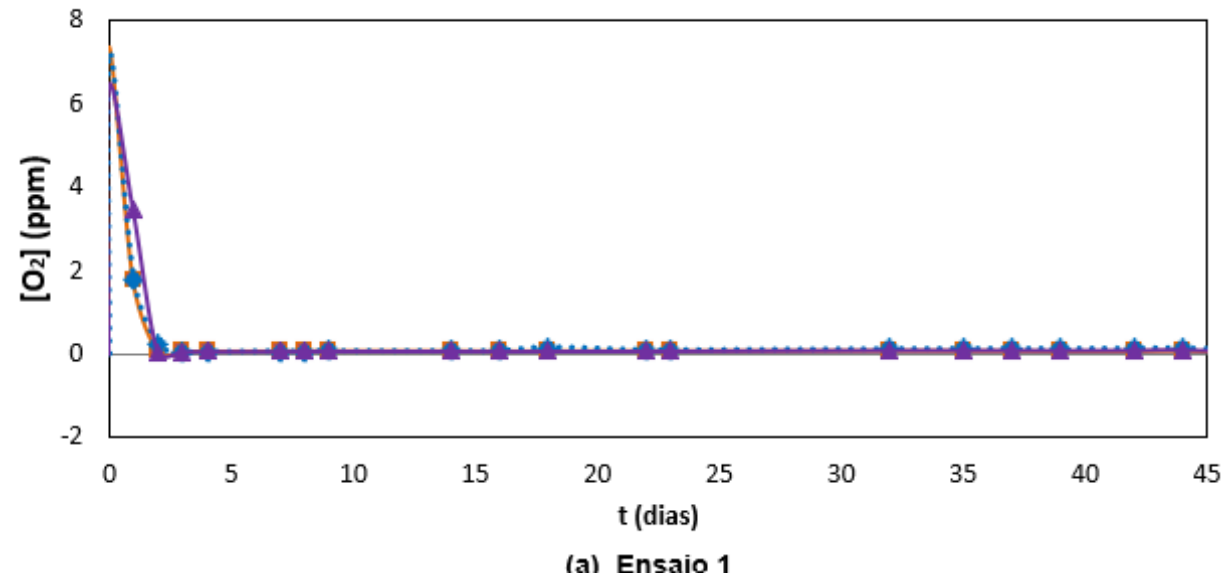
25 °C



Tests performed adopting different V/A

SAF 2205 25 °C

Natural sea water: $V/A = 1,7 \text{ mL/cm}^2$, $V/A = 6 \text{ mL/cm}^2$ e no metal



- ✓ Oxygen is consumed in a few days, regardless of the V/A ratio and even in the absence of metal, confirming once again that the rapid consumption is due to the microbial activity present in natural seawater.

No micro-organisms' growing were
detected on the test biocupons
after the tests

Conclusions



- ✓ The optical luminescence O_2 sensors proved to be quite suitable for monitoring the dissolved O_2 content at the temperatures adopted in the present study (5 °C, 15 °C and 25 °C), presenting adequate durability for tests of up to 60 days, without presenting any type of failure.
- ✓ At 25 °C, O_2 consumption is rapid in non-sterilized natural seawater and practically zero in sterilized natural and synthetic seawater, indicating that O_2 is consumed by microorganisms present in natural seawater and not due to metal passivation as cited in the literature.
 - ✓ Most likely the amount of O_2 consumed by steel passivation is so small that it is not detected by dissolved O_2 content monitoring.
- ✓ O_2 consumption decreased with lowering the temperature in all conditions studied, most likely due to the decrease in bacterial activity with temperature.

Conclusões



➤ **AISI 304**

- The critical pitting temperature of AISI 304 steel in seawater is close to 15 °C. For this reason, this steel showed significant localized corrosion on the cutting edges at 25 °C, slight localized corrosion at 15 °C, and did not show any pitting at 5 °C.
- The O₂ consumed due to localized corrosion of AISI 304 was detected by the O₂ sensors.

➤ In future studies on dissolved O₂ consumption, it is recommended that only steels that are not susceptible to localized corrosion be used.

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*Thank
you!*